

Seasonable And Spatial Deviation Of Groundwater Quality In Tannery Villages Of Ambur, Vellore

S. Vasanthan¹ and A. Murugesan²

1. Manonmaniam Sundaranar University, Centre for Research, Abishekapatti, Tirunelveli – 627 012

2. Government Arts College, P.G. and Research, Department of Chemistry, Ariyalur - 621 713

The present study regulates the richness of heavy metals (Cr, Cu, Zn, Fe, Ni, Al, Co, Mn and Pb) and study of physico-chemical parameters in the groundwater (pre and post-monsoon) of the Ambur region and estimates the pollution indices and risk assessment to assess the correctness of groundwater for human consumption. The suitability of groundwater for drinking and irrigation has been measured in north and eastern part of Vellore district. In two different seasons (pre and post-monsoon), 7 groundwater samples collected in the same location during March 2017 and October 2017 near tanneries in different villages in Ambur taluk, Vellore district, Tamil Nadu. In the majority of the groundwater samples, the concentration of heavy metals exceeds the desirable limits of Bureau of Indian Standards (BIS), 2012, whereas average Cr concentration is 0.759 and 0.675 mg/L during pre and post-monsoon season, respectively, which is above the BIS permissible limit (0.05 mg/L) perhaps due to chromium rich salts. The calculated pollution indices, namely contamination index (CI) and index of environmental risk (IER) for the heavy metals propose that the majority of the studied groundwater samples are in the highly contaminated zone. Though, all the samples close to tannery and followings fall in the highly contaminated zone indicating their unsuitability for drinking and agricultural activity purposes.

KEYWORDS

Heavy metals, Groundwater, Tannery, Ambur, Pollution indices, Risk assessment

REFERENCES

1. Boakye, E. 2008. *European J. Scientific Res.*, 21(4):617-626.
2. Meyer, W.B. and B.L. Turner. 1994. Changes in landuse and land cover : A global perspective. Cambridge University Press, Cambridge, England, New York, NY. U.S.A.
3. Sajil Kumar, P., J.P. Jegathambal and E.J. James. 2011. Multivariate and geostatistical analysis of ground water quality in Palar river basin. *Int.J. Geology*. 5(4).
4. Vennila, G., T. Subramani and L. Elango. 2008. GIS based ground water quality-Assessment of Vattamalaikarai basin, Tamil Nadu. *Int. Quarterly Scientific J.*, 7(4):585-592.
5. Doneen, L.D. 1964. Notes on water quality in agriculture. Water Science and Engineering, University of California, Davis.
6. Ganeshkumar, B. and C. Jaideep. 2012. Groundwater quality assessment using water quality index (WQI) approach-Case study in a coastal region of Tamil Nadu. *Int. J. Env. Sci. and Res.*, 1(2):50-55.
7. Hussein, H.O. 2015. Assessment of groundwater quality in some rural areas of the federal capital territory, Abuja, Nigeria. *Int. J. Scientific Res. and Education*.
8. Krishna Kumar, S., et al. 2014. Hydro geochemistry and groundwater quality appraisal of part of south Chennai coastal aquifers, Tamil Nadu using WQI and fuzzy logic method. *Appl. Water Sci.*, 13201-013-0148-4.
9. Mohapatra. 2016. Ecology of polluted water. APH Publication Co., New Delhi. pp 1-47.
10. Punia, Anita and Neelam Siva Siddaiah. 2017. *Asian J. Water Env. and Poll.*, 14(4):9-19.
11. Sadashivaiah, C., C.R. Ramakrishnaiah and G. Ranganna. 2008. Hydrochemical analysis and evaluation of groundwater quality in Tumkur taluk, Karnataka State.
12. Sadashivaiah, C., C.R. Ramakrishnaiah and G. Ranganna. 2008. *Int. J. Env. Res. Public Health*. 5(3):158-164.
13. Tiwari, T.N. and M.A. Mishra. 1985. A preliminary assignment of water quality index of major Indian rivers. *Indian J. Env. Prot.*, 5:276-279.

14. Upadhyay, A. 2014. Pre-monsoon study of physico-chemical parameters of Hemavathi river, Turuvekere, Karnataka. *Int. J. Innovative Res. in Sci. Eng. and Tech.*, 3(9):15986-15990.
15. Siddiqui, A., S. Naseem and T. Jalil. 2005. Groundwater quality assessment in and around Kalu Khuhar, super highway, Sindh, Pakistan. *J. Appl. Sci.*, 5(7):1260-1265.
16. Singh, A.K., *et al.* 2010. Quality assessment of mine water in the Raniganj Coalfield area. *Mine Water the Env.*, 29:248-262.

Removal Of Nickel From Electroplating Effluent Treatment Plant Sludge Using *Helianthus annuus* L.

M.J. Suresh and P. Rajiv

Karpagam Academy of Higher Education, Department of Biotechnology, Coimbatore - 641 021

Pollutants from the industries are a major concern rather than those from regular human activities. In particular, non-degradable pollutants and heavy metals are released into the environment through untreated wastewater and effluents. There are 70% of nickels present in electroplating sludge. The current work proves that *H. annuus* is able to remove nickel from the pretreated electroplating sludge. Different concentrations of electroplating effluent treatment plant (ETP) sludge and chelating agent (EDTA) were mixed with red soil. The pH, nitrogen, potassium, phosphate and heavy metals of raw sludge, red soil and ETP treated sludge was analysed. Phytoremediation studies were carried using *H. annuus* on pre-treated sludge. The moderate shoot, root length and dry and fresh weight of *H. annuus* were observed. Nickel level and other physico-chemical properties were analysed in post-harvested pre-treated sludge. A significant level of nickel reduction was observed from this investigation. Thus, the research work concluded that *H. annuus* was the best server in the removal of heavy metal (nickel) from the sludge.

KEYWORDS

Helianthus annuus, Nickel, Phytoremediation, Sludge

REFERENCES

1. Blaylock, M.J. and J.W. Huang. 2000. Phytoextraction of metals. In Phytoremediation of toxic metals : Using plants to clean-up the environment. Ed I. Raskin and B.D. Ensky. John Wiley and Sons, Inc., New York. pp 53-70.
2. Dembitsky, V. 2003. Natural occurrence of arseno compounds in plants, lichens, fungi, algal species, and microorganisms. *Plant Sci.*, 165:1177-1119.
3. Zhen-Guo, S., et al. 2002. Lead phytoextraction from contaminated soil with high biomass plant species. *J. Env. Qual.*, 31:1893-1900.
4. Bryan, G.W. 1980. Recent trends in research on heavy-metal contamination in the sea. *Helgolander Meeresuntersuchungen.* 33:6-25.
5. MoE. 2001. Nickel in the environment. Fact sheet. Ministry of the Environment, Ontario.
6. Hussain, S.T., T. Mahmood and S.A. Malik. 2010. Phytoremediation technologies for Ni⁺⁺ by water hyacinth. *A.J.B.* 9(50) : 8648-8660.
7. Gerendas, J., et al. 1999. Significance of nickel for plant growth and metabolism. *J. Plant Nutr. Soil Sci.*, 162:241-256.
8. Pendias, A.K. and H. Pendias. 2001. Trace elements in soils and plants (3rd edn). CRC Press, Boca Raton. pp 285.
9. Ozbek, H., et al. 1995. Soil science. Cukurova University Agricultural Faculty Publ., Adana. pp 352.
10. Davari, M. and M. Homaie. 2012. A new yield multiplicative model for simultaneous phytoextraction of Ni and Cd from contaminated soils. *J. Water Soil.* 25(6):1332-1343.
11. Hassan, Z. and M.G.M. Aarts. 2011. Opportunities and feasibilities for biotechnological improvement of Zn, Cd or Ni tolerance and accumulation in plants. *Env. Exp. Bot.*, 72(1): 53-63.
12. Martin, T.A. and M.V. Ruby. 2004. Review of in situ remediation technologies for lead, zinc and cadmium in soil. *Remed. J.*, 14:35-53.
13. Saifullah, M.E., et al. 2009. EDTA assisted Pb phytoextraction. *Chemosphere.* 74:1279-1291.
14. Panwar, B.S., K.S. Ahmed and S.B. Mittal. 2002. Phytoremediation of nickel-contaminated soils by *Brassica* species. *Env. Develop. Sustain.*, 4(1):1-6.
15. Mathur, N., et al. 2007. Arbuscular Mycorrhizal fungi: A potential tool for phytoremediation. *J. Plant Sci.*, 2:127-140.

16. Greipsson, S. 2011. Phytoremediation. *Nat. Edu. Know.* 2: 7.
17. Visoottiviset, P., K. Francesconi and W. Sridok-chan. 2002. The potential of Thai indigenous plant species for the phytoremediation of arsenic contaminated land. *Env. Poll.*, 118:453-461.
18. Parnian, A., et al. 2012. Phytoremediation of nickel from hydroponic environment using *Ceratophyllum demersum*. *Greenhouse Sci. Tech.*, 6:75-84.
19. APHA. 1989. Standard methods for examination of water and wastewater (20th edn). American Public Health Association, Washington, D.C.
20. Strickland, J.D.H. and T.R. Parsons. 1968. A practical handbook of seawater analysis. *Bull. Fish. Res. Board Canada.* 167:49-62.
21. Jansen, L.L. and E.H. Cronin. 1953. *Halogeton* on trial. *Farm and Home Sci., (Utah).* 14:38-39.
22. Raskin, I. and B.D. Ensley. 2000. Phytoremediation of toxic metals: using plants to clean up the environment. John Wiley and Sons, Inc., New York. pp 53-70.
23. Jeeva, A., et al. 2016. Effect of textile wastewater irrigation on seed germination, plant growth, biomass and crop yield in green gram seeds (*Vignaradiate (L) Wilczeż*) under plating technique and pot experiment. *J. Env. Sci., Toxicol. Food Tech.*, 10:86-89.
24. Pilon-Smits, E., 2005. Phytoremediation. *Annu. Rev. Plant. Biol.*, 5:15–39.
25. Rahman, H., et al. 2005. Effects of nickel on growth and composition of metal micronutrients in barley plants grown in nutrient solution. *J. Plant Nut.*, 28:393-404.
26. Chhotu, D., M. Jadia and H. Fulekar. 2008. Phytoremediation: The application of vermicompost to remove zinc, cadmium, copper, nickel and lead by sunflower plant. *Env. Eng. Manage. J.*, 7:547–558.
27. Seregin, I.V. and A.D. Kozhevnikova. 2006. Physiological role of nickel and its toxic effects on higher plants. *Rus. J. Plant Phy.*, 53: 257- 277.
28. Alkorta, I., et al. 2004. Recent findings on the phytoremediation of soils contaminated with environmentally toxic heavy metals and metalloids, such as zinc, cadmium, lead, and arsenic. *Rev. Env. Sci. Biotech.*, 1 : 71- 90.
29. Panda, S.K. and S. Choudhury. 2005. Chromium stress in plants. *Braz.J. Plant Physiol.*, 17:131-136.

Diagnostic Approach Of Air Pollution Using Statistical Prediction Analysis: Case Of Kenitra City

Hadine Mohsine¹, Bahi Lahcen¹, Azami Rajaa², Arfaoui Amine³ and Rachidai Abdelatif⁴

1. *Mohammadia School of Engineers (EMI), Applied Geophysics Laboratory, The Engineering Geology and Environment, Agdal, Rabat, Morocco*

2. *Mohammadia School of Engineers (EMI), Applied Quality of Air and the Water Laboratory, The Section of Environment, Agdal, Rabat, Morocco*

3. *Royal Institute of Managers Training, Department of Life Sciences, Sale, Morocco*

4. *University Ibn Tofail, Kenitra, Morocco*

The present study focuses on the air quality of Kenitra city (northwest of Morocco). Eight sites with different conditions were chosen. The targeted parameters were: the suspended particulate matter (SPM), lead (Pb), cadmium (Cd), nitrogen dioxide (NO₂), sulphur dioxide (SO₂) which represent the independent variables (IV) and the vehicular traffic intensity which represents the dependent variable (DV). In addition to the evaluation of concentrations of each pollutant in the studied sites, correlations between the explanatory factors (IV) and the DV were calculated. The diagnosis of the used approach- multiple linear regression, Pearson correlation, significance level, analysis of variance (ANOVA), correlation coefficient R, the value of (t) and the determination coefficient R². The results demonstrated that NO₂ is strongly correlated with traffic intensity, consequently, it could be considered as a pollution indicator of urban air linked to vehicle traffic. The interest of such analysis and interpretation methodology lies in the possibility to predict and, therefore, to prevent air pollution by targeting the means required to reduce or stop it.

KEYWORDS

Air quality, Characterization, Pollution, Suspended particulate matter, Pb, Cd, NO₂, SO₂, Multiple regression, Correlation, Indicator, Kenitra, Morocco.

REFERENCES

1. Katz, M., *et al.* 1970. Measuring air pollutants : A guide to choosing methods. World Health Organization, Geneva. 130 pp.
2. Afnor, 1996. Air quality environment. pp 35-38.
3. Afnor, 1991. Gas analysis. Air quality (vol 1). pp 178-179.
4. Reynald, Lev. 2007. SPSS programming and data management : A guide for SPSS Inc., Chicago Ill.
5. Breimain, L. and J.H. Friedman. 1985. Estimating optimal transformations for multiple regression and correlation. *J. Am. Statistical Assoc.*, 80:580-598.
6. Labreuche, J. 2010. Less differents types de variables, leurs repre sentations graphiques et parameters descriptifs. *Sang Thrombose Vaisseaux*. 22:536-543.
7. Fabienne, C., M. Corneau and M.M. Cousineau. 2010. Guided introduction au logiciel SPSS. CRI-1600 G : Initiation aux methods quantitatives. Certificate de criminologie. Faculte de l'education permanente. pp 66-70.
8. Falissard, B. 1988. Comprendre et utiliser les. statistiques dans les sciences de la vie. Masson, Paris. pp 65-69.
9. Nyberg, F., *et al.* 2000. Urban air pollution and lung cancer in stocklom. *Epidemiology*. 11:487-495.
10. Nfstad, P., *et al.* 2003. Lung cancer and air pollution : A 27 years follow-up of 16 209 Norwegian men. *Thorax*. 58:1071-1076.
11. Tenenhaus, M. 2007. Methods pour decrire, expliquer et previor, collection : Management sub dunod. Statistique (2nd edn). pp 38-42.

12. Dodge, Y. and V. Rousson. 2004. Analysis of applied regression. Collection : Eco sup, Dunod. pp 56-62.

Assessment Of Heavy Metal Contamination And Its Effect On Colonial Wetland Birds - A Review

Bibhu Prasad Panda¹, Abanti Pradhan², Siba Prasad Parida³ and Aditya Kishore Dash²

1. Siksha 'O' Anusandhan (Deemed to be University), Environmental Sciences, Institute of Technical Education and Research, Bhubaneswar - 751 030

2. Siksha 'O' Anusandhan (Deemed to be University), Biofuel and Bioprocessing Research Center, Institute of Technical Education and Research, Bhubaneswar - 751 030

3. Centurion University of Technology and Management, Department of Zoology, School of Applied Sciences, Bhubaneswar- 752 050

The aquatic and terrestrial ecosystems are in threat due to metal pollution and this is the hub of disease for wild species existence. Birds are the most sensitive organism to the environmental contaminants than other vertebrates. Heavy metals enter into the body mainly by food chain and accumulated in various body organs and transferred to the offspring. When the metal concentration exceeds the limit, it harms the organism causing reproduction failure and also kidney toxicity due to the increasing concentration of Zn. The widespread birds, like egrets and herons are used as bioindicators or biomonitors for local contamination due to their high position in the food chain. In the present review, various organs, such as eggs, liver, blood, feathers, eggshells of the birds have been used to estimate heavy metal accumulation. Water birds face many health implications due to the higher concentration of heavy metals in water. The embryogenesis, hatching success and viability is affected by chromium, lead and cadmium. Different concentrations of metals are found in different tissues of the body, so the study of different tissues is necessary for the population level evaluation. It is suggested that the use of feathers are the non-destructive way for assessing the heavy metal contamination. Concentrations of heavy metals in feathers serve as proof of the heavy metal circulation in the blood. Hence, it is very important to assess the heavy metal accumulation from the feathers of the colonial water birds as the effective heavy metal bioindicator.

KEYWORDS

Colonial wetland birds, Heavy metal, Feathers, Review.

REFERENCES

1. Qadir, A., R.N. Malik and S.Z. Husain. 2008. Spatio temporal variations in water quality of Nullah Aik-tributary of the river Chenab, Pakistan. *Env. Monit. Assess.*, 140:43-59.doi:10.1007/s10661-007-9846-4.
2. Qadir, A. and R.N. Malik. 2009. Assessment of an index of biological integrity (IBI) to quantify the quality of two tributaries of river Chenab, Sialkot, Pakistan. *Hydrobiologia*. 621:127-153.
3. Covaci, A., *et al.* 2002. Hair analysis : Another approach for the assessment of human exposure to selected persistent organochlorine pollutants. *Chemosp.*, 46:413-418. doi:10.1016/S0045-6535(01)00065-0.
4. Mateo, R. and R. Guitart. 2003. Heavy metals in livers of water birds from Spain. *Arch. Env. Contam. Toxicol.*, 44:398-404.doi:10.1007/s00244-002-2040-3.
5. Dauwe, T., *et al.* 2003. Variation of heavy metals within and among feathers of birds of prey : Effects of molt and external contamination. *Env. Poll.*, 124:429-436.doi:10.1016/S0269-7491(03)0004-7.
6. Dauwe, T., *et al.* 2004. Relationships between metal concentrations in great tit nestlings and their environment of food. *Env. Poll.*, 131:373-380. doi:10.1016/j. envpol.2004.03.009.
7. Martion-Diaz, M.L., *et al.* 2005. Bioaccumulation and toxicity of dissolved heavy metals from the Guadalquivir estuary after the Aznalcollar mining spill using *Ruditapes philippinarum*. *Arch. Env. Contam. Toxicol.*, 48:233-241. doi:10.1007/s00244-003-9202-9.

8. Eeva, T., E. Belskii and B. Kuranov. 2006. Environmental pollution affects genetic diversity in wild bird populations. *Mutat. Res.*, 608:8-15.
9. Dural, M., *et al.* 2006. Bioaccumulation of some heavy metals in different tissues of *Dicentrarchus labrax* 1, 1758, *Sparus aurata* 1, 1758 and *Mugilcephalus* 1, 1758 from the c, Amlık lagoon of the eastern coast of Mediterranean (Turkey). *Env. Monit. Assess.*, 118:65-74. doi:10.1007/s10661-006-0987-7.
10. Kojaddinovic, J., *et al.* 2007. Trace elements in three marine birds breeding on Reunion Island (Western Indian Ocean) : Part 2-Factors influencing their detoxification. *Arch. Env. Contam. Toxicol.*, 52:431-440. doi:10.1007/s00244-005-8225-9.
11. Ullah, K., M.Z. Hashmi and R.N. Malik. 2014. Heavy-metal levels in feathers of cattle egret and their surrounding environment : A case of the Punjab Province, Pakistan. *Arch. Env. Contam and Toxicol.*, 66:139-153.
12. De Luca-Abbott, S.B., *et al.* 2001. Review of effects of water pollution on the breeding success of water birds, with particular reference to ardeids in Hong Kong. *Ecotoxcol.*, 10:327-329.doi: 0.1023/A:1012288811808.
13. Eqani, S., *et al.* 2012. Distribution and risk assessment of organochlorine contaminants in surface water from river Chenab, Pakistan. *J. Env. Monit.*, 14(6):1645-1654.
14. Gochfeld, M. 1997. Spatial patterns in a bioindicator heavy metal and selenium concentration in eggs of Herring Gulls (*Larus argentatus*) in the New York Bight. *Env. Contam. Toxicol.*, 33:63-70.
15. Furness, R. 1993. **In** Birds as monitors of environmental change. **Ed** R.W. Furness and J.J.D. Greenwood. Chapman and Hall, London.
16. Zhang, Y., *et al.* 2006a. Little egret (*Egretta gazetta*) and trace metals concentration in wetland of China. *Env. Monit. Assess.*, 118:355-368.doi.10.1007/s10661-006-1496-4.
17. Zhang, Y., *et al.* 2006b. Little egrets (*Egretta garzetta*) and trace metal contamination in wetlands of China. *Env. Monit. Assess.*, 118(1):355-368.
18. Kim, J. and T.H. Koo. 2008a. Heavy metal concentrations in feathers of Korean shorebirds. *Arch. Env. Contam. Toxicol.*, 55:122-128.doi:10.1007/s00244-007-9089-y.
19. Kim, J. and T.H. Koo. 2008b. Heavy metal distribution in chicks of two heron species from Korea. *Arch. Env. Contam. Toxicol.*, 54(4):740-747.
20. Scheuhammer, A.M. 1987. The chronic toxicity of aluminium, cadmium, mercury and lead in birds : A review. *Env. Poll.*, 46:263-265.
21. Pereira, M.G., *et al.* 2009. Long-term trends in mercury and PCB congener concentrations in gannet (*Morus bassanus*) eggs in Britain. *Env. Poll.*, 157:155-163.
22. O' Halloran, I., *et al.* 2003. Mercury and organochlorine content of Dipper *Cincluscinclus* eggs in south-west Ireland : Trends during 1990-1999. *Env Poll.*, 123(1):85-93.
23. Malik, R.N. and N. Zeb. 2009. Assessment of environmental contamination using feathers of *Bubulcus ibis* L. as a biomonitor of heavy metal pollution, Pakistan. *Ecotoxicology*. 18(5):522-536.
24. Eens, M., *et al.* 1999. Great and blue tits as indicators of heavy metal contamination in terrestrial ecosystems. *Ecotoxicol. Env. Saft.*, 44:81-85. doi:10.1006/eesa.1999.1828.
25. Ayas, Z. 2007. Trace element residues in eggshells of grey heron (*Ardeacinerea*) and black-crowned night heron (*Nycticoraxnycticorax*) from Nallihan Bird Paradise, Ankara, Turkey. *Ecotoxicol.*, 16:347-352.doi:10.1007/s10646-007-0132-6.
26. Dauwe, T., *et al.* 2005. The reproductive success and quality of blue tits (*Parus ceruleans*) in a heavy metal pollution gradient. *Env. Poll.*, 136:243-251.doi:10.1016/j.envpol.2005.01.009.
27. Shahbaz, M., *et al.* 2013. Relationship between heavy metals concentrations in egret species, their environment and food chain differences from two wetlands of Pakistan. *Chemosphere*. 93:274-282.
28. Burger, J. 1994. Metals in avian feathers : Bioindi-cators of environmental pollution. *Rev. Env. Toxicol.*, 5:203-211.
29. Burger, J. 1995. Heavy metal and selenium levels in feathers of herring gulls (*Larus argentalus*). Differences due to years, gender and age at capture.

30. Scheifler, R., *et al.* 2006. Lead concentrations in feathers and blood of common blackbirds (*Turdus merula*) and in earthworms inhabiting unpolluted and moderately polluted urban areas. *Sci. Total Env.*, 371:197-205.
31. Burger, J., *et al.* 2007. Mercury, arsenic, cadmium, chromium, lead and selenium in feathers of pigeon guillemots (*Cephus columba*) from Prince William Sound and Aleutian Islands of Alaska. *Sci. Total Env.*, 387:175-184.
32. Deng, H., *et al.* 2007. Trace metal concentration in great tit (*Parus major*) and greenfinch (*Carduelis sinica*) at the western mountains of Beijing, China. *Env. Poll.*, 148:620-626.doi:10.1016/j.envpol.2006.11.012.
33. Horai, S., *et al.* 2007. Trace element accumulations in 13 avian species collected from Kanto area, Japan. *Sci. Total Env.*, 373:512-525.doi:10.1016/j.scitotenv.2006.10.010.
34. Burger, J. and M. Gochfeld. 1993. Heavy metal and selenium levels in feathers of young egrets and herons from Hong Kong and Szechuan, China. *Arch. Env. Contam. Toxicol.*, 25:322-327.doi:10.1007/BF0021074.
35. Fasola, M., *et al.* 1998. Heavy metals, organo-chlorine pesticides and PCBs in eggs and feather of heron breeding in northern Italy. *Arch. Env. Contam. Toxicol.*, 34:87-93.doi:10.1007/s002449900289.
36. Movalli, P.A. 2000. Heavy metal and other residues in feathers of Laggar falcon *Falco biarmicus jugger* from six districts of Pakistan. *Env. Poll.*, 109:267-275. doi:10.1016/S0269-7491(99) 00258-4.
37. Lucia, M. 2010. Effects of dietary cadmium contamination on bird *Anas platyrhynchos*–Comparison with species *Cairina moschata*. *Ecotoxicol. Env. Safe.*, 73(8):2010-2016.
38. Weyers, B., E. Gluck and M. Stoepler. 1988. Investigation of the significance of heavy metal contents of blackbirds feathers. *Sci. Total Env.*, 77:61-67.doi:10.1016/0048-9697(88)90315-4.
39. Kim, J. and T.H. Koo. 2007a. Heavy metal concentrations in diet and livers of black-crowned night heron *Nycticorax nycticorax* and grey heron *Ardeacitiera* chicks from Pyeongtaek, Korea. *Ecotoxicol.*, 16:411-416.doi:10.1007/s10646-007-0143-3.
40. Burger, J. and M. Gochfeld 2000c. Effects of lead on larids : A review of laboratory and field studies. *J. Toxicol. Env. Health.* 3:59-78.doi:10.1080/109374000281096.
41. Goutner, V., *et al.* 2001. Lead and cadmium in eggs of colonially nesting water birds of different position in the food chain of Greek wetlands of international importance. *Sci. Total Env.*, 267:169-176.doi:10.1016/S0048-9697(00)00808-1.
42. Malik, R.N. and S.Z. Husain. 2006a. Spatial distribution of ecological communities using remotely sensed data. *Pak. J. Bot.*, 38:571-582.
43. Malik, R.N. and S.Z. Husain. 2006b. Land-cover mapping : A remote sensing approach. *Pak J. Bot.*, 38:559-570.
44. Malik, R.N. and S.Z. Husain. 2006c. Classification and ordination of vegetation communities of the Lohibehr reserve forest and its surrounding areas, Rawalpindi, Pakistan. *Pak. J. Bot.*, 38:543-558.
45. Malik, R.N. and S.Z. Husain. 2007. *Broussonetia papyrifera* (L) L'He'r. ex Vent : An environmental constraint on the Himalayan foot-hills vegetation. *Pak. J. Bot.*, 39:1045-1053.
46. Kim, J. and T.H. Koo. 2007 b. The use of feathers to monitor heavy metal contamination in herons, Korea. *Arch. Env. Contam. Toxicol.*, 53:435-441.doi:10.1007/s00244-006-0196-y.
47. Burger, J. and M. Gochfeld. 2004a. Metal levels in eggs of common terns (*Sterna hirundo*) in New Jersey : Temporal trends from 1971 to 2002. *Env. Res.*, 94:336-343. doi:10.1016/S0013-9351(03) 00081-1.
48. Burger, J. 2008. Assessment and management of risk to wildlife from cadmium. *Sci. Total Env.*, 389(1):37-45.
49. Burger, J. and M. Gochfeld. 2000b. Metals in albatross feathers from Midway Atoll : Influence of species, age and nest location. *Env. Res.*, 82:207-221.doi:10.1016/enrs.1999.4015.
50. Johansen, P., *et al.* 2004. High human exposure to lead through consumption of birds hunted with lead shot. *Env. Poll.*, 127(1):125-129.

51. Hui, C.A. 2002. Concentrations of chromium, manganese and lead in air and in avian eggs. *Env. Poll.*, 120:201-206.doi:10.1016/S0296-7491(02) 00158-6.
52. Janssens, T.K.S., *et al.* 2009. Molecular mechanisms of heavy metals tolerance and evolution in invertebrates. *Insect. Sci.*, 16(1):3-18.
53. Mico, C., *et al.* 2006. Assessing heavy metal sources in agricultural soils of an European Mediterranean area by multivariate analysis. *Chemosphere*. 65(5):863-872.
54. Wei, B.G. and L.S. Yang. 2010. A review of heavy metal contaminations in urban soils, urban road dusts and agricultural soils from China. *Microchem. J.*, 94(2):99-107.
55. Syed, J.H. and R.N. Malik. 2011. Occurrence and source identification of organochlorine pesticides in the surrounding surface soils of the Ittehad Chemical Industries, Kalashah Kabu, Pakistan. *Env. Earth Sci.*, 62(6):1311-1321.
56. Pain, D.J., *et al.* 1999. Levels of polychlorinated biphenyls, organochlorine pesticides, mercury and lead in relation to shell thickness in marsh harrier (*Circus aeruginosus*) eggs from Charente-Maritime, France. *Env. Poll.*, 104:61-68. doi:10.1016/S0269-7491(98)00154-7.
57. Tariq, S.R., *et al.* 2008. Statistical source identification of metals in groundwater exposed to industrial contamination. *Env. Monit. Assess.*, 138(1-3):159-165.
58. Hansen, B.H., *et al.* 2007. Induction and activity of oxidative stress related proteins during waterborne Cd/Zn-exposure in brown trout. (*Salmo trutta*). *Chemosphere*. 67(11):2241-2249.
59. Huang, S.S., *et al.* 2007. Survey of heavy metal pollution and assessment of agricultural soil in Yangzhong district, Jiangsu Province, China. *Chemosphere*. 67(11):2148-2155.
60. Dauwe, T., *et al.* 2006. Effects of heavy metal exposure on the condition and health of adult great tits (*Parus major*). *Env. Poll.*, 140:71-78. Doi:10.1016/jenvpol.2005.06.024.
61. Spahn, S.A. and T.W. Sherry. 1999. Cadmium and lead exposure associated with reduced growth rates, poorer fledging success of little blue heron chicks (*Egretta caerulea*). *Arch. Env. Contam. Toxicol.*, 37:377-384.
62. Carpenter, J.W., G.A. Andrews and W.N. Beyer. 2004. Zinc toxicosis in a free-flying trumpeter swan (*Cygnus buccinator*). *J. Wild Dis.*, 40(4):769-774.
63. Bostan, N., *et al.* 2007. Diagnosis of heavy metal contamination in agro-ecology of Gujranwala, Pakistan using cattle egret (*Bubulcus ibis*) as bioindicator. *Ecotoxicology*. 16(2):247-251.
64. Burger, J., *et al.* 1992. Heavy metal and selenium levels in young cattle egrets from nesting colonies in the northeastern United States, Puerto Rico and Egypt. *Arch. Env. Contam. Toxicol.*, 23:435-439.
65. Burger, J. 1993. Metals in avian feathers : bioindicators of environmental pollution. *Rev. Env. Toxicol.*, 5:203-311.
66. Takekawa, J.Y., *et al.* 2002. Relating body condition to inorganic contaminant concentrations of diving ducks wintering in coastal California. *Arch. Env. Cont. and Toxi.*, 42:60-70.
67. Kerte'sz, V. and T. Fa'ncsi.2003. Adverse effects of (surface water pollutants) Cd, Cr and Pb on the embryogenesis of the mallard. *Aqua. Toxicol.*, 65:425-433.doi:10.1016/S0166-445x(03)00155-3.
68. Pan, C., *et al.* 2008. Concentrations of metals in liver, muscle and feathers of tree sparrow : Age, inter-clutch variability, gender and species differences. *Bull. Env. Contam. Toxicol.*, 81(6):558-560.
69. Bickham, J.W., *et al.* 2000. Effects of chemical contaminants on genetic diversity in natural populations : Implications for biomonitoring and ecotoxicology. *Mutat. Res.*, 463:33-51.doi:10.1016/s1383-5742(00)00004-1.
70. Dahl, C.R., *et al.* 2001. Cytochrome b sequences in black-crowned night-herons (*Nycticorax nycti corax*) from heronries exposed to genotoxic contaminants. *Ecotoxicol.*, 10:291-297. doi:10.1023/A:1016711401809.
71. Burger, J. and M. Gochfeld. 2000a. Effects of lead on birds (Laridae) : A review of laboratory and field studies. *J. Toxicol. Env. Health. Crit. Rev.*, 3(2):59-78.
72. Pilastro, A., *et al.* 1993. The use of bird feathers for monitoring of cadmium pollution. *Arch. Env. Contam. Toxicol.*, 24:355-358.doi:10.1007/BFO1128733.
73. Nam, D.H., D.P. Lee and T.H. Koo. 2003. Monitoring for lead pollution using feathers of feral pigeons (*Columba livia*) from Korea. *Env. Monit. Assess.*, 95:13-22.doi:10.1023/B:EMAS.0000029898.28393.30.

74. Jaspers, V.L.B., *et al.* 2007. Evaluation of the usefulness of bird feathers as a non-destructive biomonitoring tool for organic pollutants : A comparative and meta-analytical approach. *Env. Int.*, 33:328-337.
75. Van den Steen, E., *et al.* 2007. Experimental evaluation of the usefulness of feathers as a nondestructive biomonitor for poly-chlorinated biphenyls (PCBs) using silastic implants as a novel method of exposure. *Env. Int.*, 33:257-264.doi:10.1016/j.envint. 2006.09.018.
76. Thyen, S., P.H. Becker and H. Behmann. 2000. Organochlorine and mercury contamination of little terns (*Sterna albifrons*) breeding at the western Baltic sea, 1978-96. *Env. Poll.*, 108:225-238.
77. Custer, T.W., *et al.* 2007. Selenium and metal concentrations in water-bird eggs and chicks at Agassiz National Wildlife Refuge, Minnesota. *Arch. Env. Contam. Toxicol.*, 53(1):103-109.
78. Hashmi, M.Z., R.N. Malik and M. Shahbaz. 2012. Heavy metals in eggshells of cattle egret (*Babulcus ibis*) and little egret (*Egretta garzetta*) from the Punjab province, Pakistan. *Ecotoxicol. Env. Safe.*, 89:158-165.
79. Pappas, A.C., *et al.* 2006. Interspecies variation in yolk selenium concentrations among eggs of free-living birds : The effect of phylogeny. *J. Trace. Elem. Med. Biol.*, 20:155-160. doi:10.1016/j.jtemb.2006.03.001.
80. Dauwe, T., *et al.* 2004b. The effect of heavy metal exposure on egg size, eggshell thickness and the number of spermatozoa in blue tit *Parus caeruleus* eggs. *Env. Poll.*, 129(1):125-129.
81. Rattner, B.A., *et al.* 2008. Concentrations of metals in blood and feathers of nestling ospreys (*Pandion haliaetus*) in Chesapeake and Delaware Bays. *Arch. Env. Contam. Toxicol.*, 54:114-122.doi:10.1007/s00244-007-9004-6.
82. Nam, D.H., *et al.* 2005. Specific accumulation of 20 trace elements in great cormorants (*Phalacrocorax carbo*) from Japan. *Env. Poll.*, 134:503-514.doi:10.1016/j.envpol.2004.09.003.
83. Roger, F., *et al.* 2005. Mercury concentrations in muscle brain and bone of western Alaskan water bird. *Sci. Total Env.*, 349:277-283.
84. Goodale, M.W., *et al.* 2008. Marine foraging birds as bioindicators of mercury in the Gulf of Marine. *Eco. Health.* 5:409-425.
85. Cid, F.C., *et al.* 2009. Contamination of heavy metals in birds from Embalse La Florida (San Luis, Argentina). *J. Env. Monit.*, 11(11):2044.2051.
86. Jaspers, V., *et al.* 2004. The importance of exogenous contamination on heavy metal levels in bird feathers. A field experiment with free-living great tits, *Parus major*. *J. Env. Mon.*, 6(4):356-360.
87. Dauwe, T., *et al.* 2002. Great and blue tit feathers as biomonitors for heavy metal pollution. *Ecol. Indic.*, 1:227-234.doi:10.1016/S1470-160x(02) 00008-0.
88. Boncompagni, E., *et al.* Egrets as monitors of trace metal contamination in wetlands of Pakistan. *Arch. Env. Contam. Toxicol.*, 45:399-406.doi:10.1007/s00244-003-0198-y.
89. Burger, J. 1996. Heavy metal and selenium levels in feathers of Franklin's gulls in interior North America. *Auk.*, 113:399-407.
90. Burger, J. and M. Gochfeld. 1997. Heavy metal and selenium concentration in feathers of egrets from Bali and Sulawesi Indonesia. *Arch. Env. Contam. Toxicol.*, 32:217-221. doi:10.1007/s002449900178.
91. Guo, D.L., *et al.* 2001. Preliminary studies on the level and distribution of mercury in feathers of birds. *Acta. Zoologica. Sinica.* 47:139-149.
92. Burger, J. and M. Gochfeld. 2000d. Metal levels in feathers of 12 species of seabirds from Midway Atoll in the northern Pacific Ocean. *Sci. Total Env.*, 257:37-52.doi:10.1016/S0048-9697(00) 00496-4.
93. Veerle, J., *et al.* 2004. The importance of exogenous contamination on heavy metal levels in bird feathers. A field experiment with free-living great tits, *Parus major*. *J. Env. Monit.*, 6:356-360.doi: 10.1039/b314919f.
94. Dmowski, K. 1999. Birds as bioindicators of heavy metal pollution : Review and examples concerning European species. *Acta Ornithol.*, 34:1-25.

95. Manjula, M., *et al.* 2015. Biomonitoring of heavy metals in feathers of eleven common bird species in urban and rural environments of Tiruchirappalli. *Env. Monit. Assess.*, 187:267.
96. Abbasi, N.A., *et al.* 2015. Spatial and interspecific variation of accumulated trace metals between remote and urbane dwelling birds of Pakistan. *Ecotoxicol and Env. Saft.*, 113:279-286.
97. Burger, J. and M. Gochfeld. 2008. Mercury and other metals in feathers of common eider (*Somateria mollissima*) and tufted puffin (*Fratercula cirrhata*) from the Aleutian Chain of Alaska. *Arch. Env. Contam. Toxicol.*, 56:596-606.doi:10.1007/s00244-008-9207-5.

Water Quality Index To Assess Suitability Of Groundwater For Drinking In The Coastal Regions Of East Godavari District

Poosalayya Sangadi and Chandrasekar Kuppan

Vignan's Foundation for Science, Technology and Research (VFSTR), Division of Chemistry, Department of Science and Humanities, Vadlamudi, Guntur - 522 213

Water quality index (WQI) is one of the most efficient tools to assess water quality. This study was carried out by subjecting groundwater samples from 82 villages of 12 mandals of coastal habitations in East Godavari district of Andhra Pradesh. Samples were collected during post-monsoon season and analysed for physico-chemical parameters, such as pH, total alkalinity, total hardness, calcium hardness, magnesium hardness, chloride, nitrate, fluoride, total dissolved salts and electrical conductivity. Results showed that almost all parameters are exceeding the permissible limits prescribed by standards of BIS, ICMR and WHO. The WQI values range from 116.6 to 27.4. The computed WQI shows that only 17 sampling sites come under good water category followed by 27, 32 and 6 sampling sites for poor, very poor and unsuitable water categories, respectively. Majority of the samples (79.3%) were found to be unsuitable for drinking for the people residing in and around the respective sampling site. The data from this work will be very much useful for the governing bodies and scientific authorities to take immediate action in controlling the impurities in these places by implementing modern and efficient techniques to purify the water and make them portable for the common people.

KEYWORDS

Groundwater quality, Physico-chemical parameters, Coastal habitations

REFERENCES

1. Gangavarapu, Surekha, Udaya Bhaskar Pinnamaneni and Padma Kumari. 2015. Groundwater quality mapping of East Godavari district, Andhra Pradesh, using remote sensing and geostatistic. *Int. J. Recent Scientific Res.*, 6(4): 3602-3608.
2. Saxena and Saxena. 2015. The statistical assessment of fluoride and nitrate contamination status of ground water in various tehsil of Jaipur district, Rajasthan. *Int. J. Res. Studies in Biosci.*, (IJRSB). 3(3):107-131.
3. Saxena, U. and S. Saxena. 2013. Statistical assessment of ground water quality using physico-chemical parameters in Bassi tehsil of Jaipur district, Rajasthan. *Global J. Sci. Frontier Res. Env. and Earth Sci.*, 13(3):Version 1.0.
4. Gopalkrushna, Murhekar. 2011. Determination of physico-chemical parameters of surface water samples in and around Akot city. *Int. J. Res. Chem. Env.*, 1(2):183-187.
5. Kinzelbach, W. 1986. Groundwater modeling : An introduction with sample programmes in BASIC, Elsevier, New York. 333 pp.
6. Chauhan, A. and S. Singh. 2010. Evaluation of Ganga water for drinking water for drinking purpose by water quality index at Rishikesh, Uttarakhand. *Report and Opinion*. 2(9):53-61.
7. Brown, E., M.W. Skougstad and M.J. Fishman. 1974. Method for collection and analysis of water sample for dissolved minerals and gases (book no. 5). U.S. Department of Interior, Washington, D.C.
8. APHA. 2005. Standard methods for the examination of water and wastewater (21st edn). American Public Health Association, Washington, D.C.
9. Hussain, H.M., *et al.* 2014. Evaluation and mapping groundwater suitability for irrigation using GIS in Najaf Governorate, Iraq. *J. Env. Hydrology*. 22:4-14.
10. Brown, R.M., *et al.* 1972. A water quality index-crossing the psychological barrier. International Conference on Water pollution research. Jerusalem. Proceedings, 6:787-797.
11. Saxena, Swati, Umesh Saxena and A.K. Sinha. 2015. Seasonal and spatial variation of water quality index of Bassi tehsil of district Jaipur, Rajasthan. *Global J. Sci. Frontier Res. : B. Chemistry*. 15(6): Version 1.0.

12. Tyagi, Shweta, *et al.* 2013. Water quality assessment in terms of water quality index. *Am. J. Water Resour.*, 1(3):34-38.
13. Bouslah, Soraya, Lakhdar Djemili and Larbi Houichi. 2017. Water quality index assessment of Koudi at Medouar reservoir north east Algeria using weighted arithmetic index method. *J. Water and Land Develop.*, 35(10-12):221-228.
14. Rao, N.S. 2006. Seasonal variation of groundwater quality in a part of Guntur district, Andhra Pradesh. *Env. Geol.*, 49:413-429.
15. Mitharwal, S., R.D. Yadav and R.C. Angasaria. 2009. Water quality analysis in Pilani of Jhunjhunu district, Rajasthan. *Rasayan J. Chem.*, 2(4):920-923.
16. Totawat, R.K. and C.P. Singh Chandel. 2007. Quality of groundwater of Jaipur city, Rajasthan and its suitability for domestic and irrigation purpose. *Appl. Ecology and Env. Res.*, 6(2):79-88.

Performance Evaluation Of *Moringa oleifera* Seeds For Softening Of Surface And Groundwater

V.T. Gaikwad¹ and G.R. Munavalli²

1. Padmabhoosan Vasantraodada Patil Institute of Technology, Department of Civil Engineering, Sangli-416 304
2. Walchand College of Engineering, Department of Civil Engineering, Sangli - 416 415

The softening study was conducted using naturally dried *Moringa oleifera* seed kernel extract, which was found to possess some softening potential. The study was carried out using synthetic water samples of calcium, magnesium and combinations of calcium and magnesium solutions. The *Moringa oleifera* strength study revealed that 0.5-1% extract is suitable and effective for softening. The optimum pH range was found to be 5-8. *Moringa oleifera* seeds were found effective for calcium removal (removal efficiency upto 30%) but are not effective for magnesium removal.

KEYWORDS

Calcium, Hardness, Magnesium, *Moringa oleifera*, Softening

REFERENCES

1. Sani, M.A. 1990. The use of Zogale seeds for water treatment. B. Eng., Final Year Project. Bayero University, Kano, Nigeria.
2. Muyibi, A. Suleyman and Lilian M. Evison. 1994. *Moringa oleifera* seeds for softening hard water. *Water Res.*, 29(4):1099-1105.
3. Varekar V.B., 2009. Water softening by seed extracts. M.Tech. Dissertation. Department of Civil Engineering, Walchand College of Engineering, Sangli.
4. Ghanem, Ana V., James C. Young and Findly G. Edwards. 2007. Mechanisms of ballasted flocculation. *J. Env. Eng., Am. Soc. for Civil Eng.*, 133(3): 271-277.
5. Bourdages, Gaetan, Christian Scott and Charles D. Blumenschein. 2009. Innovative municipal water softening using high-rate sand-ballasted technology. *Env. Sci. and Eng. Magazine.* 20-29.
6. Mangale, Sapana M., Sonal G. Chonde and P.D. Rout. 2012. Use of *Moringa oleifera* (drum stick) seeds as natural adsorbent and an antimicrobial agent for groundwater treatment. *Res. J. Recent Sci.*, 1(3):31-30.
7. Bell-Ajy, K., *et al.* 2000. Conventional and optimized coagulation for NOM removal. *J. Am. Water Works Assoc.*, 92:44-57.
8. APHA, AWWA, WEF. 1998. Standard methods for the examination of water and wastewater (20th end). American Public Health Association, American Water Works Association, Water Environment Federation, Washington, D.C.

Adsorption Potential Of Water Hyacinth On Removal Of Direct Blue 2 Dye From Aqueous Solution: Isotherm, Kinetics And Thermodynamics Studies

J. Padmapriya¹, G. Vijayaraghavan^{1,2} and S. Shanthakumar¹

1. Vellore Institute of Technology (VIT), Department of Environmental and Water Resources Engineering, School of Civil Engineering, Vellore - 632 014

2. Rajalakshmi Engineering College, Department of Chemical Engineering, Thandalam, Chennai - 602 105

The adsorption potential of water hyacinth on the removal of direct blue 2 (DB2) dye from aqueous solution was investigated. Surface area, morphology and functional groups were identified using Brunauer, Emmett and Teller (BET), scanning electron microscopy (SEM) and Fourier transform infrared (FTIR) analysis, respectively. All the essential process influencing parameters, such as pH (2-4), initial dye concentration (100-300 mg/L), adsorbent dosage (1-5 g/L), contact time of adsorbent with pollutant (15-180 min) and temperature (20-40°C) were studied. The maximum dye removal efficiency of 87.8% was observed at pH 2, the adsorbent dosage of 3 g/L, initial dye concentration of 100 mg/L, 40°C temperature and contact time of 90 min. The experimental isotherm data were analyzed using the Langmuir, Freundlich and Temkin models. Equilibrium and kinetic studies revealed that the Freundlich isotherm and pseudo second order kinetic model were the best fit for the adsorption process. The thermodynamic analysis showed that the adsorption process is spontaneous and endothermic in nature.

KEYWORDS

Adsorption, Water hyacinth, Direct blue 2 dye, Kinetics, Isotherm

REFERENCES

1. Akar, S.T., *et al.* 2009. Biosorption of a reactive textile dye from aqueous solutions utilizing in agro-waste. *Desalination*. 249(2):757-761.
2. Hajati, S., *et al.* 2014. Competitive adsorption of direct yellow 12 and reactive orange 12 on ZnS : Mn nanoparticles loaded on activated carbon as novel adsorbent. *Ind. Eng. Chem.*, 20(2):564-571.
3. Sulak, M.T., E. Demirbas and M. Kobya. 2007. Removal of astrazon yellow 7GL from aqueous solutions by adsorption onto wheat bran. *Bioresour. Tech.*, 98(13):2590-2598.
4. Karmaker, S., *et al.* 2015. Adsorption of reactive orange 13 onto jackfruit seed flakes in aqueous solution. *J. Env. Chem. Eng.*, 3(1):583-592.
5. Ahmad, M.A., A. NurAzree and S.B. Olugbenga. 2014. Kinetic, equilibrium and thermodynamic studies of synthetic dye removal using pomegranate peel activated carbon prepared by microwave-induced KOH activation. *Water Resour. Ind.*, 6(1):18-35.
6. Mittal, A., J. Damodar and J. Mittal. 2013. Adsorption of hazardous dye Eosin yellow from aqueous solution onto waste material de-oiled soya : Isotherm kinetics and bulk removal. *J. Mol. Liq.*, 179(1):133-140.
7. Ozturk, A. and E. Malkoe. 2014. Adsorptive potential of cationic basic yellow 2 (BY2) dye onto natural untreated clay (NUC) from aqueous phase : Mass transfer analysis, kinetic and equilibrium profile. *Appl. Surf. Sci.*, 299:105-115.
8. Djilani, C., *et al.* 2015. Adsorption of dyes on activated carbon prepared from a pricot stones and commercial activated carbon. *J. Taiwan Inst. Chem. Eng.*, 53:1-10.
9. Namasivayan, C. and D. Kavitha. 2002. Removal of congo red from coir pith and agricultural solid waste. *Dyes Pigm.*, 54:47-58.
10. Auta, M. and B.H. Hameed. 2011. Preparation of waste tea activated carbon using potassium acetate as an activating agent for adsorption of acid blue 25 dye. *Chem. Eng. J.*, 171(2):502-509.

11. Zhang, R., *et al.* 2014. Adsorption of congo red from aqueous solutions using cationic surfactant modified wheat straw in batch mode : Kinetic and equilibrium study. *J. Taiwan Inst. Chem. Eng.*, 45(5):2578-2583.
12. Zhao, B., *et al.* 2014. Adsorption of light green anionic dye using cationic surfactant-modified peanut husk in batch mode. *Arabian J. Chemistry*. 10(S2):S3595-S3602.
13. Ardejani, P.D., *et al.* 2008. Adsorption of direct red 80 dye from aqueous solution onto almond shells : Effect of pH, initial concentration and shell type. *J. Hazard. Mater.*, 151:730-737.
14. Attia, A.A., W.E. Rashwan and S.A. Khedr. 2006. Capacity of activated carbon in the removal of acid dyes subsequent to its thermal treatment. *Dyes Pigm.*, 69:128-136.
15. Villamagna, A. and B. Murphy. 2010. Ecological and socio-economic impacts of invasive water hyacinth (*Eichhornia crassipes*) : A review. *Freshw. Biol.*, 55(2):282-298.
16. Patel, S. 2012. Threats, management and envisaged utilizations of aquatic weed *Eichhornia crassipes* : An overview. *Rev. Env. Sci. Biotech.*, 11(11):249-259.
17. Padamapriya, J. and S. Shanthakumar. 2016. Removal of cationic dye (basic green) using raw and modified *Eichhornia crassipes* : Adsorption kinetics, isotherms, thermodynamics and mechanism studies. *Indian J. Env. Prot.*, 36(12):1001-1004.
18. Prola, L.D., *et al.* 2013. Adsorption of direct blue 53 dye from aqueous solutions by multi-walled carbon nanotubes and activated carbon. *J. Env. Manage.*, 130:166-175.
19. Silvalacerda, V.D., *et al.* 2015. Rhodamine B removal with activated carbons obtained from lignocellulosic waste. *J. Env. Manage.*, 155:67-76.
20. Ghaedi, M., *et al.* 2012. Kinetics, thermodynamics and equilibrium evaluation of direct yellow 12 removal by adsorption onto silver nanoparticles loaded activated carbon. *Chem. Env. J.*, 187:133-141.
21. Vucurovic, V.M., *et al.* 2014. Removal of anionic azo dyes from aqueous solutions by adsorption on maize stem tissue. *J. Taiwan Inst. Chem. Eng.*, 45(4):1700-1708.
22. Pavia, D.L., G.M. Lampman and G.S. Kriz. 2004. Introduction to spectroscopy and infrared spectroscopy. Thomson Asia Pvt. Ltd., Singapore. pp 92.
23. Kuo, C.Y., C.H. Wu and J.W. Wu. 2008. Adsorption of direct dyes from aqueous solutions by carbon nanotubes : Determination of equilibrium, kinetics and thermodynamics parameters. *J. Colloid Interface Sci.*, 372(2):308-315.
24. Xi, Y., *et al.* 2013. Removal of azo dye from aqueous solution by a new biosorbent prepared with *Aspergillus nidulans* cultured in tobacco waste-water. *J. Taiwan Inst. Chem. Eng.*, 44(5):815-820.
25. Konicki, W., *et al.* 2012. Adsorption of anionic dye direct red 23 onto magnetic multi-walled carbon nanotubes-Fe₃C nano-composite : Kinetics, equilibrium and thermodynamics. *Chem. Eng. J.*, 210:87-95.
26. Machado, F.M., *et al.* 2011. Adsorption of reactive red M-2B E dye from water solutions by multi-walled carbon nanotubes and activated carbon. *J. Hazard. Mater.*, 192:1122-1131.
27. Fathi, M.R., A. Asfaram and A. Farhangi. 2015. Removal of direct red 23 from aqueous solution using corn stalks : Isotherms, kinetics and thermodynamic studies. *Spectrochim. Acta A Mol. Biomol. Spectrosc.*, 135:364-372.
28. Liu, S., *et al.* 2014. Adsorption of the anionic dye congo red from aqueous solution onto natural zeolites modified with N, N-dimethyl dehydroable-tylamine oxide. *Chem. Eng. J.*, 248:135-144.
29. Sayed, El. Z. and El. Ashtoukhy. 2009. *Loofa egyptiaca* as a novel adsorbent for removal of direct blue dye from aqueous solution. *J. Env. Manage.*, 90(8):2755-2761.
30. Amin, N.K. 2009. Removal of direct blue-106 dye from aqueous solution using new activated carbons developed from pomegranate peel : Adsorption equilibrium and kinetics. *J. Hazard. Mater.*, 165(1-3):52-62.
31. Isah, U., *et al.* 2015. Kinetics, equilibrium and thermodynamics studies of C.I reactive blue 19 dye adsorption on coconut shell based activated carbon. *Int. Biodeterior. Biodegradation*. 102:265-273.
32. Vanaamudan, A. and P.P. Sudhakar. 2015. Equilibrium, kinetics and thermodynamic study on adsorption of reactive red blue-nanoclay (Cloisite 30B) nanobiocomposite. *J. Taiwan Inst. Chem. Eng.*, 55:145-151.

Temple Ponds As A Traditional Water Management Option In Kerala

G. Lakshmi

Cochin University of Science and Technology, School of Environmental Studies, Kochi - 682 022

Kerala, the southern state of India is blessed with water resources which include a number of rivers, lakes and small ponds. Kerala is also a land of many small and large temples and it has been found that even from the primitive times, water resource conservation and management was considered as a matter of great importance. Temple ponds and tanks in Kerala are visible examples of the community initiative for water management through cultural involvement. Most of these ponds are conserved by the native people of an area, giving them religious respect. But these water bodies still faces problems, like oxygen depletion, eutrophication and pollution. To reduce these problems and to enhance the water quality in the temple tanks and ponds, certain management measures need to be adopted.

KEYWORDS

Deterioration, Ecosystem, Eutrophication, Pollution, Protection, Rainwater harvesting, Religious, Sustainable, Traditional, Water quality

REFERENCES

1. Pandey, D.N. 2005. Traditional knowledge systems for biodiversity conservation. Available at https://www.infinityfoundation.com/mandala/t_es/t_es_pande_conserve.htm.
2. Pandey, D.N., A.K. Gupta and D.M. Anderson. 2003. Rainwater harvesting as an adaptation to climate change. *Curr. Sci.*, 85(1):46-59.
3. Das, K. and S. Gupta. 2012. Seasonal variation of Hemiptera community of a temple pond of Cachar district, Assam. *J. Threatened Tara.*, 3050-3058.
4. Sreenivasan, A. 1964. The limnology, primary production and fish production in a tropical pond, *Limnol. Oceanogr.*, 9(3):391-396.
5. Sreenivasan, A. 1976. Limnological studies of and primary production in temple pond ecosystems. *Hydrobiologia.* 48(2):117-123.
6. Arulmurugan, P., S. Nagaraj and N. Anand. 2010. Biodiversity of fresh water algae from temple tanks of Kerala. *Rec. Res. Sci. Tech.*, 2(6):58-71.
7. Ajayan, K.V., M. Selvaraju and K. Thirugnana-moorthy. 2013. Phytoplankton population of Ananthapura temple lake of Kasaragod, Kerala. *Insight Bot.*, 3(1):6-14.
8. Jipsa, J.R., *et al.* 2013. Physico-chemical and biological study on the lentic water bodies of Palakkad, Kerala, *IOSR J. Pharmacy and Biological Sci.*, 8(1): 21-25.
9. Paul, P.T. and P.K. Anu. 2016. Algal diversity of Guruvayur temple pond, Thrissur district, Kerala, *Int. J. Advanced Life Sci.*, 9(3):302-306.
10. Amsha Devi, V., S. Baskaran and R. Suresh Kumar. 2013. Physico-chemical parameters and zooplankton diversity of a temple pond in Virudhunagar, Tamil Nadu. *Int. J. Sci. Env. Tech.*, 2(2):250-257.
11. Ragi, M.S. and D.S. Jaya. 2014. Distribution and diversity of oligochaetes in selected ponds of Thiruvananthapuram district, Kerala. *Advances in Ecology.* doi:10.1155/2014/138360.
12. Ravichandran, C., S. Suthabala and S. Jayalakshmi. 2009. Environmental quality of selected temple ponds in Tiruchirappalli, *Indian J. Env. Prot.*, 29 (5): 392-398.
13. Sulabha, V. and V. R. Prakasam. 2006. Limnological features of Thirumullavaram temple pond of Kollam Municipality, Kerala. *J. Env. Biol.*, 27(2): 449-451.
14. Thirugnanamoorthy, K. and M. Selvaraju. 2009. Phytoplankton diversity in relation to physico-chemical parameters of Gnanaprekasam temple pond of Chidambaram in Tamil Nadu. *Recent Res. in Sci. and Tech.*, 1(5):235-238.
15. Baruah, P.P. and B. Kakati. 2013. Water quality and phytoplankton diversity of Gopeswar temple freshwater pond in Assam. *Bangl. J. Bot.*, 41(2): 181-185.

16. Ganesh, K.S., *et al.* 2015. Physico-chemical properties of Sri Sanishwaran temple pond water, Tirunallar. *World Scientific News*. 21:1-11.
17. Narayan, R., K.K. Saxena and S.Chauhan. 2007. Limnological investigations of Texi temple pond in district Etawah (U.P.). *J. Env. Biol.*, 28(1):155.

Real Time And Wireless Groundwater Level Recharging And Monitoring System In Borewell

X. Anitha Mary¹, Lina Rose¹, K. Rajasekaran¹ and S. Asokan²

1. *Karunya Institute of Technology and Sciences, Coimbatore*

2. *SALZER Electronics Limited, Coimbatore*

Water is a primary resource and plays a significant role in human existence. The rate at which the water being used is exponentially increasing day by day. All living organisms need water for their survival. About 98% of the Earth's fresh water is groundwater. There are many water storage phenomenon where the major among them is groundwater. As water is consumed, naturally subsequent amount of water will be added to the aquifer to reinforce human needs. This recharge capacity if measured will give a new eye-opening to those areas which are facing severe drought conditions. The groundwater level should be monitored continuously, serves essential to predict the water resource availability for the future. So, it's necessary to log the data at regular intervals of time in a database, thus enabling us to check and predict the area scarce of water. A pressure sensor helps in finding the water level. The physical pressure is converted to the level or height of the water column concerned parameter. The water level measurements are acquired during particular intervals of time and are viewed as HTML data.

KEYWORDS

Borewell, Water level, Microcontroller, Pressure sensor, XBee, Mobile app

REFERENCES

1. Bartram, Jamie and Richard Balance. 1996. Water quality monitoring : A practical guide to the design and implementation of freshwater quality studies and monitoring programmes. CRC Press
2. Khaled Reza, S.M., Shah Ahsanuzzaman Md. Tariq and S.M. Mohsin Reza. 2010. Microcontroller based automated water level sensing and controlling: Design and implementation issue. World Congress on Engineering and computer science. Proceedings, pp 220-224.
3. Lee, B.Y. and B.Y. Park. 1999. Development of high precision underground water level meter using a buoyant rod load cell technique. *KSAFM*. 1:1-5.
4. Rodriguez, Francisco M. Lopez. 2016. Low cost educational mobile robot based on android arduino. *J. Intelligent and Robotic Systems*.
5. Lim, Seong Peng and Gik Hong Yeap. 2011. Centralised smart home control system via Xbee transceivers. Humanities, Sci. and Eng., (CHUSER). IEEE Colloquium.
6. Shajahan, Altaf Hamed and A. Anand. 2013. Data acquisition and control using arduino-android platform-smart plug. IEEE, ICEETS.
7. Raspberry Pi. Hardware information, http://elinux.org/RPi_Hardware.
8. How SoC works. <http://www.androidauthority.com/how-it-works-systems-on-a-chip-soc-93587/>
9. Boot process. <http://thekandyancode.wordpress.com/2013/09/21/how-the-raspberry-pi-boots-up/>.
10. NOOBS. <http://www.raspberrypi.org/introducing-noobs/>.

Trend Analysis Of Rainfall Over Bihar Region

Shraddha Yadav, A.K. Sachan and D. Basu

Motilal Nehru National Institute of Technology, Department of Civil Engineering, Allahabad - 211 004

The present study aims to examine the long-term changes and short-term fluctuations in annual and seasonal rainfall over the Bihar region of India for the period of 1954-2013. The analysis was carried out to identify trends in the precipitation pattern. It is clear from these results that rainfall yield is increasing in pre-monsoon (MAM) season and monsoon season (JJAS) and the anomaly is (+3.9) mm in decades V, VI, (+4.8) mm in VI decade. In J-F and OND season, decreasing rainfall trend after IV decade is observed. Periodicity is 9-11 years suns pot number (SSN) long term and 2.5 years quasi-biennial oscillation (QBO) short term periodicity observed in wavelet analysis, these relationships are noticed in rainfall by using cross wavelet analysis. Breaks and active periods are identified during the southwest monsoon season (June–September) and active days identified in recent extreme rainfall year in the period July-August month, maximum active and break days in the year 2007 and 2010. Outgoing longwave radiation (OLR) positive anomaly increases when rainfall decreases (break days) in 2010 and when OLR negative anomaly increases with increased rainfall (active days) in 2007. It is also evident that the Indian summer monsoon extremity is strongly associated with equatorial Indian Ocean oscillation (EQUINOO), which is based on the surface zonal wind over the central equatorial Indian Ocean.

KEYWORDS

Rainfall, Cramer test, Standardized anomaly index, Wavelet transform, Active and break days

REFERENCES

1. Jones, P.D. and M. Hulme. 1996. Calculating regional climatic series for temperature and precipitation. *Int. J. Climatology*. 16:361-377.
2. Gadgil, S., *et al.* 2004. Extremes of the Indian summer monsoon rainfall, ENSO and equatorial Indian Ocean oscillation. *Geophys. Res. Lett.*, 31:L12213.
3. Hiremath, K.M. and P.I. Mandi. 2004. Influence of the solar activity on the Indian monsoon rainfall. *New Astronomy*. 651-662.
4. Ramamurthy, K. 1969. Monsoon of India : Some aspects of the 'break' in the Indian southwest monsoon during July and August. Forecasting manual no. IV. Indian Meteorological Department, Pune. pp 18.3:1-57.
5. Krishnamurthy, V. and J. Shukla. 2008. Intraseasonal persistence and propagation of intraseasonal patterns over the Indian summer monsoon region. *Climate Dynamics*. 30:353-369.
6. Abaje, I.B. and P.N. Giwa. 2010. Flood risk assessment and vulnerability in Kafanchan town, Jema'a local Government area of Kaduna State, Nigeria. *Int. J. Sustainable Develop.*, 3(1):94-100.
7. Torrence, C. and G.P. Compo. 1997. A practical guide to wavelet analysis. *Bull. Am. Meteor. Soc.*, 79(1):61-78.
8. Bhattacharya, S. and R. Narasimha. 2005. Possible association between Indian monsoon rainfall and solar activity.

Groundwater Characterization And Treatment Using UV/H₂O₂ Process

Jatinder Kumar Ratan

Dr. B. R. Ambedkar National Institute of Technology, Department of Chemical Engineering, Jalandhar-144 011

The present study has a connection with the real-life problem of colour and contamination in the groundwater of the Dasuya town in the state of Punjab. An assessment of the groundwater contamination has been carried out in the laboratory. The physico-chemical parameters have been determined. Elemental analysis has been performed using microwave plasma atomic emission spectrometer. The organic compounds present in the groundwater were identified by gas chromatography and mass spectroscopy analysis. Many treatment technologies have been tested for treatment of the organic contamination present in the groundwater. It was observed that the UV/H₂O₂ process showed the potential to treat contaminants of the groundwater. An annular reactor has been used to treat the groundwater by UV/H₂O₂ process. A parametric study has been performed to optimize the dose of H₂O₂ and pH for treatment of the groundwater.

KEYWORDS

UV/H₂O₂ process, Annular reactor, Groundwater treatment, Advanced oxidation processes, Physico chemical parameters

REFERENCES

1. Fernando, R., *et al.* 2015. Analysis of trace elements in ground water using ICP-OES and TXRF techniques and its compliance with Brazilian Protection Standards. *Water Air Soil Poll.*, 226:32.
2. Arjoon, A., A.O. Olaniran and B. Pillay. 2013. Co-contamination of water with chlorinated hydrocarbons and heavy metals : Challenges and current bioremediation strategies. *Int. J. Env. Sci. Tech.*, 10:395-412.
3. Gonza'lez-Naranjo, V. and K. Boltes. 2013. Toxicity of ibuprofen and perfluorooctanoic acid for risk assessment of mixtures in aquatic and terrestrial environments. *Int. J. Env. Sci. Tech.*, 11:1743-750.
4. Taghipour, H., *et al.* 2013. Heavy metals pollution in the soils of suburban areas in big cities : A case study. *Int. J. Env. Sci. Tech.*, 10:243-250.
5. Katukiza, A.Y., *et al.* 2015. Grey water characterisation and pollutant loads in an urban slum. *Int. J. Env. Sci. Tech.*, 12:423-436.
6. Kumar, J. and A. Bansal. 2012. Photocatalytic degradation of amaranth in aqueous solution catalyzed by immobilized nanoparticles of titanium dioxide. *Int. J. Env. Sci. Tech.*, 9:479-484.
7. Natarajan, T.S., *et al.* 2013. Study on identification of leather industry wastewater constituents and its photocatalytic treatment. *Int. J. Env. Sci. Tech.*, 10:855-864.
8. Tunay, O., *et al.* 1996. Colour removal from textile wastewaters. *Water Sci. Tech.*, 34:9-16.
9. Slokar, Y.M. and A.M.L. Marechal. 1998. Methods of decolouration of textile wastewaters. *Dyes Pigments.* 37:356.
10. Galindo, C., P. Jacques and C. Kalt. 2001. Photooxidation of the phenylazonaphthol A020 on TiO₂ : Kinetic and mechanistic investigations. *Chemosphere.* 45:997-1005.
11. Daneshvar, N., *et al.* 2006. Photocatalytic degradation of the herbicide erioglaucine in the presence of nanosized titanium dioxide : Comparison and modeling of reaction kinetics. *J. Env. Sci. Health B.* 41:1273-1290.
12. Toor, A.P., *et al.* 2006. Photocatalytic degradation of direct yellow 12 dye using UV/TiO₂ in a shallow pond slurry reactor. *Dyes Pigments.* 68:53-60.

13. Natarajan, T.S., *et al.* 2011. Study on UV-LED/TiO₂ process for degradation of rhodamine B dye. *Chem. Eng. J.*, 169:126-134.
14. Kumar, J. and A. Bansal. 2013. A comparative study of immobilization techniques for photocatalytic degradation of Rhodamine B using nanoparticles of titanium dioxide. *Water Air Soil Poll.*, 224:1452.
15. Oppenlander, T. 2009. Photochemical purification of water and air. Wiley-VCH, Weinheim.
16. Kitano, M., *et al.* 2007. Recent developments in titanium oxide-based photocatalysts. *Appl. Catal. A*. 325:1-14.
17. Poyatos, J., *et al.* 2009. Advanced oxidation processes for wastewater treatment : State of the art. *Water Air Soil Poll.*, 205:187-204.
18. Sano, T., *et al.* 2008. Degradation of C₂H₂ with modified-TiO₂ photocatalysts under visible light irradiation. *J. Mol. Catal. A*. 284:127-133.
19. Kumar, J. and A. Bansal. 2013. Photocatalytic degradation in annular reactor : Modelization and optimization using computational fluid dynamics (CFD) and response surface methodology (RSM). *J. Env. Chem. Eng.*, 1:398-405.
20. Elmolla, E.S. and M. Chaudhari. 2015. Comparison of different advanced oxidation processes for treatment of antibiotic aqueous solution. *Desalination*. 256:43-47.
21. Tuhkanen, T.A. 2004. Advanced oxidation processes for water and wastewater treatment. IWA, London.
22. Liao, C.H. and M.D. Gurol. 1995. Chemical oxidation by photolytic decomposition of hydrogen peroxide. *Env. Sci. Tech.*, 29:3007-3014.
23. APHA. 2005. Standard methods for the examination of water and wastewater. American Public Health Association, Washington.
24. Assadi, A. and A. Eslami. 2010. Comparison of phenol photodegradation by UV/H₂O₂ and photo-fenton process. *Env. Eng. Manage. J.*, 9:807-812.
25. Kavitha, V. and K. Polanivelu. 2005. Destruction of cresols by fenton oxidation process. *Water Resour.*, 39:3062-3072.

Design And Implementation Of Stable PID Controller For A Quadruple Tank System

P. SubhaHency Jose, J. Jerome Thomas and P. Rajalakshmy

Karunya University, Department of Electronics and Instrumentation Engineering, Coimbatore - 641 114

In many industries, the interaction process is taken only in single loop controls where the multi-input and multi-output (MIMO) systems are difficult to control in the water treatment process. Thus, the proportional integral derivative (PID) controller is used to control the liquid level by adjusting the controller parameters and the variables of flow and level of four tanks. The quadruple tank system is a four interconnected water tank system used to analyze the linear and non-linear effect of interaction process taken in chemical industries and water treatment plant. The design of stable PID controller is done with stability region analysis method for the minimum and non-minimum phase of the tank level provides better disturbance rejection and good setpoint tracking and simulation is done. The MATLAB/Simulink and LabVIEW is used to simulate the stable PID controller and to analyze the stability region.

KEYWORDS

Multi-input and multi-output (MIMO), Proportional integral derivative (PID), Quadruple tank system, Pulse width modulation (PWM), Ziegler-Nichols method

REFERENCES

1. Firouzbahrami, Majid and Amin Nobakhti. 2016. Cement on design of PID controllers for interval plants with time delay. *J. Process Control*. 44:160-161.
2. Ghousiya Begum, K., A. Seshagiri Rao and T. K. Radhakrishnan. 2017. Enhanced IMC based PID controller design for non-minimum phase (NMP) integrating processes with time delays. *ISA Trans.*, 68:223-234.
3. Kadu, C. B. and C. Y. Patil. 2016. Design and implementation of stable PID controller for interacting level control system. 7th International Conference on Comm., computing and virtualization. *Procedia Comp. Sci.*, 79:737-746.
4. Malar, R. Suja Mani and T. Thyagarajan. 2009. Modeling of quadruple tank system using soft computing tech., *Eu. J. Sci. Res.*, 29(1-2):249.
5. Malwatkar, G.M., A.A. Khandekar and S.D. Nikam. 2014. PID controllers for higher order systems based on maximum sensitivity function. *Int. J. Electronics Comp. Tech.*
6. Roinila, Tomi, Matti Vilkkko and Antti. 2008. Corrected mathematical model of quadruple tank process. 17th World Congress. The Int. Federation of Automatic Control.
7. Rosinova, Danica and Matus. 2008. Robust control of quadruple tank process. *ICIC Exp. Letters ICIC Int.*, 2(3).
8. Saeed Qamar, Vali Uddin and Reza Katebi. 2010. Multivariable predictive PID control for quadruple tank. *World Academy of Sci., Eng. and Tech.*, pp 861-866.
9. Srivastava, Saurabh, et al. 2016. An optimal PID controller via LQR for standard second order plus time delay systems. *ISA J. Trans.*, 60:244-253.

Solar Residential Rooftop Systems Business Models: SWOT Analysis

Akhil Sarin^{1,2}, Rahul Gupta¹ and Vishwajeet V. Jituri³

1. Amity University, Amity Business School, Noida

2. WREL - A Tata Power Company Limited, Noida

3. Toshiba India Pvt. Ltd., Gurugram

The carbon footprint on earth due to the burning of fossil fuels for generating electricity, running trains, cars and other vehicles throughout the world is increasing day by day. It has reached the stage where if not managed effectively, it would become, like a black hole and suck the life of human beings on this earth hardly in next 50 years. Therefore, the whole world is looking for renewable sources, preferably solar energy to generate power, which is clean with no fuel cost and available in abundance in most of the countries across the globe including India. India has set a target of installation of 100 GW (including 40 GW of solar rooftop systems) of solar power plants by 2022. Residential roofs have huge installation potential and, therefore, it is one of the major target segments for the installation of solar rooftop systems. Presently, CAPEX and RESCO are the two major solar residential rooftop systems (SRRSs) business models existing in the market. Both the models are prevalent and the customers have the choice to go for any of the models for getting SRRS installed on their residential roofs. In order to enable various stakeholders involved in SRRSs, like roof owners, third party, developer, engineering, procurement and construction (EPC) firm to take an informed decision about the selection of a suitable model for particular installations, there is a need to understand the strengths and weaknesses of both the models. At present, SWOT analysis of CAPEX and RESCO models is not readily available and there is visible gap existing in its literature. Therefore, this study has been undertaken for conducting a SWOT analysis of both the business models (CAPEX and RESCO) to help stakeholders identify and adopt a suitable model for respective installations.

KEYWORDS

Solar energy, Solar residential rooftop systems (SRRSs), SWOT analysis, Business models, Renewable energy, Solar policy, Environment

REFERENCES

1. CEA. 2018. Power sector-April 2018. Central Electricity Authority, Ministry of Power, Government of India, New Delhi.
2. MNRE. 2015. Scaling up of grid connected solar power projects from 20,000 MW by the year 2021-22 to 1,00,000 MW by the year 2021-22 under National Solar mission. Ministry of New and Renewable Energy, Government of India, New Delhi.
3. MNRE. 2016. Grid connected rooftop and small solar power plants programme - Scaling up of budget from Rs. 600 crore during the 12th five year plan to Rs.5000 crore for implementation over a period of five years upto 2019-20 under National Solar Mission (NSM) - 2nd amnd. Retrieved from Ministry of New and Renewable Energy, Government of India, New Delhi. <https://mnre.gov.in/file-manager/UserFiles/gcrt-cfa-notification-02-09-2016.pdf>.
4. Sarin, A. 2009. Corporate strategic motivation: Evolution continues—Henry. A. Murray's manifest needs to Maslow's Hierarchy of needs to Anil Sarin's contributory theory of existence. *The J. Am. Academy of Business*. 14(2):237-244.
5. Frantzis, L., *et al.* 2008. Photovoltaics business models. National Renewable Energy Laboratory.
6. Lorenz, P., D Pinner and T. Seitz. 2008. The economics of solar power. *The McKinsey Quarterly*.
7. Sarin, A., R. Gupta and V. V. Jituri. 2018. Solar residential rooftop systems (SRRS) in south. *Int. J. Renewable Energy Res.*, 8(2):954-963.

8. Srivastava, S. P. and S.P. Srivastava. 2013. Solar Energy and its future role in Indian economy. *Int. J. Env. Sci.,: Development and Monitoring*. 4(3): 81-88.
9. Meena, R. S., J. S. Rathore and S. Johri. 2014. Grid connected roof top solar power generation: A review. *Int. J. Eng. Develop. and Res.*, 3(1):325-330.
10. IFC. 2014. Harnessing energy from the sun: Empowering rooftop owners. International Finance Corporation, New Delhi.
11. STA. 2015. Business model report. Solar Trade Association - PV Financing.
12. Partnerships IQ. 2015. Rooftop solar public-private partnerships : Lessons from Gujarat solar. The World Bank Group.
13. Sarin, A., and R. Gupta. 2017. Solar residential rooftop systems (SRRSs) in India : Pest and environment analysis. *Int. J. Appl. Business and Economic Res.*, 71-84.
14. Jena, L. P., *et al.* 2016. Rooftop solar private sector financing facility. The India Innovation Lab for Green Finance.
15. Goel, M. 2016. Solar rooftop in India : Policies, challenges and outlook. *Green Energy and Env.*, 1(2):129-137.
16. ACE-D. 2016. While paper on gross metering for solar rooftops in Karnataka. Patanership of Advance Clena Energy Development.
17. Burger, S. P. and M. Luke. 2016. Business models for distributed energy resources: A review and empirical analysis. MIT Energy Initiative.
18. MNRE. 2016. Performance based incentive scheme for DISCOMs for expeditious development of grid connected rooftop solar power plants. Retrieved from SPIN - An online application for solar photovoltaic installation (grid-connected rooftop). Ministry of New and Renewable Energy, Government of India, New Delhi. <https://solarrooftop.gov.in/notification/Notification-08112016901.pdf>
19. Bridge to India. 2017. India solar handbook. Bridge to India, New Delhi.
20. Castellanos, S., D. A. Sunter and D.M. Kammen, 2017. Rooftop solar photovoltaic potential in cities: How scalable are assessment approaches? *Env. Res. Letters*. 1-6. doi:<https://doi.org/10.1088/1748-9326/aa7857>.
21. Wallach, P. 2017. Leveling the playing field for rooftop solar. Center for Effective Public Management at Brookings.
22. Kotak, Y., *et al.* (n.d.). Installation of rooftop solar PV modules and their impact on building cooling load. OASYS South Asia Research Project. doi:10.1177/0143624414527098.

