

Evaluation Of Low Density Polyethylene Degradation By Bacteria Isolated From Hydrocarbon Contaminated Site

Shailja Singh¹, Vishal Singh Chowdhary¹, Devendra Pratap Singh¹, Shiv Shankar² and Shikha^{1*}

1. Babasaheb Bhimrao Ambedkar University, Department of Environmental Science, School of Earth and Environmental Sciences, Lucknow - 226 025, U.P., India

2. Gautam Buddha University, Department of Environmental Science, School of Vocational Studies and Applied Sciences, Gautam Buddh Nagar - 201 312, U.P., India

*Corresponding author, Email : envscibbau@ymail.com; shailhasingh578@gmail.com

Indiscriminate use and inertness of synthetic polymers leading to increased water and land pollution are of great concern. Many attempts have been made to control the problem by using both chemical and biological methods. Chemical methods resulted in an increase in pollution by releasing noxious gases in the atmosphere whereas biological methods have been proved to be eco-friendly. The current study demonstrates the isolation, characterization and biodegradation analysis of the low-density polyethylene (LDPE) films by the isolated bacteria. Various techniques were used to determine the changes induced in the polyethylene sheet after treatment with bacteria. Gradual reduction, in the weight of degraded LDPE with time, signifies the utilization of polymer by bacteria for nutrients and energy. Analysis through scanning electron microscopy (SEM) of the degraded LDPE films showed morphological damage, like cracks, extensive roughening, fragileness and fragmentations on the surface of the LDPE sheets confirming that degradation had occurred by the action of bacterial isolates. Fourier transform infrared spectroscopy (FTIR) analysis of the degraded LDPE films, showed the presence of alcohols, alkenes, alkanes, amines being produced after 90 days, indicating that degradation had been carried out successfully. Energy dispersive x-ray study revealed the decrease in carbon content which shows that the carbon has been utilized by bacteria for their growth. Hence, the two potential bacterial strains isolated in the present study can be said to be plastic degrading microorganisms.

KEYWORDS

Low-density polyethylene, Scanning electron microscopy, Fourier transforms spectroscopy, Biodegradation, Pollution

REFERENCES

1. Browne, M.A., et al. 2008. Ingested microscopic plastic translocates to the circulatory system of the mussel, *Mytilus edulis* (L.). *Env. Sci. Tech.*, 42(13): 5026-5031.
2. Avio, C.G., S. Gorbi and F. Regoli. 2015. Experimental development of a new protocol for extraction and characterization of microplastics in fish tissues: First observations in commercial species from Adriatic sea. *Marine Env. Res.*, 111: 18-26.
3. Hakkarinen, M. and A. Albertsson. 2004. Environmental degradation of polyethylene. *Adv. Polymer Sci.*, 169:177-199.
4. Sivan, A., M. Szanto and V. Pavlov. 2006. Biofilm development of the polyethylene-degrading bacterium *Rhodococcus ruber*. *Appl. Microbiol. Biotech.*, 72: 346-352.
5. Vijaya, C. and R.M. Reddy. 2008. Impact of soil composting using municipal solid waste on biodegradation of plastics. *Indian J. Biotech.*, 7(2): 235-239.
6. Arutchelvi, J., et al. 2008. Biodegradation of polyethylene and polypropylene. *Indian J. Biotech.*, 7:9-22.
7. Satlewal, A., et al. 2008. Comparative biodegradation of HDPE and LDPE using an indigenously developed microbial consortium. *J. Microbiol. Biotech.*, 18(3): 477-82.
8. Kowalczyk, A., et al. 2016. *Achromobacter xylosoxidans* as a new microorganism strain colonizing high-density polyethylene as a key step to its biodegradation. *Env. Sci. Poll. Res.*, 23(11): 11349-11356.
9. Bhone, M.K., et al. 2010. Biodegradation of low density polythene (LDPE) by *Pseudomonas* species. *Indian J. Microbiol.*, 52(3):411-419.

10. Das, M.P. and S. Kumar. 2013. Influence of cell surface hydrophobicity in colonization and biofilm formation on LDPE biodegradation. *Int. J. Pharmacy Pharmaceutical Sci.*, 5(4):690-694.
11. Driymal, P., J. Hoffmann and M. Druz. 2007. Evaluating the aerobic biodegradability of plastics in soil environments through GC and IR analysis of gaseous phase. *Polymer Testing*. 26(6):729-741.
12. Zerbi, G., *et al.* 1989. Structural depth profiling in polyethylene films by multiple internal reflection infrared spectroscopy. *Polymer*. 30(12): 2324-2327.
13. Awasthi, S., *et al.* 2017. Biodegradation of thermally treated low density polyethylene by fungus *Rhizopus oryzae* NS 5. *3 Biotech.*, 7(1):73.
14. Gilan, I., Y. Hadar and A Sivan. 2004. Colonization, biofilm formation and biodegradation of poly ethylene by a strain of *Rhodococcus ruber*. *Appl. Microbiol. Biotech.*, 65(1):97-104.
15. Silverstein, R.M., F.X. Webster and D.J. Kiemle. 2005. Spectrometric identification of organic compounds (7th edn). John Wiley & Sons, Inc., United States.
16. Albertsson, A.C., S.O. Andersson and S. Karlsson. 1987. The mechanism of biodegradation of polyethylene. *Polymer Degradation Stability*. 18(1): 73-87.
17. Koutny, M., J. Lemaire and A.M. Delort. 2006. Biodegradation of polyethylene films with prooxidant additives. *Chemosphere*. 64(8):1243-1252.
18. Albertsson, A.C. and S. Karlsson. 1993. Aspects of biodeterioration of inert and degradable polymers. *Int. Biodeterioration Biodegradation*. 31(3): 161-170.
19. Tribedi, P. and A.K. Sil. 2013. Low-density polyethylene degradation by *Pseudomonas* sp. AKS2 biofilm. *Env. Sci. Poll. Res.*, 20(6):4146-4153.
20. Das, M.P. and S. Kumar. 2015. An approach to low-density polyethylene biodegradation by *Bacillus amyloliquefaciens*. *3 Biotech.*, 5(1):81-86.

Wadi Larbaa's Water Quality In The North Of Morocco: Statistical Treatment Of Physical And Chemical Parameters

K. Arouya^{1,2}, H. Tabyaoui², H. Taouil¹, J. Naoura² and S. Ibn Ahmed^{1*}

1. *University Ibn-Tofail, Laboratory Materials, Electrochemistry and Environment, Faculty of Sciences, Morocco*

2. *University Sidi Mohamed Ben Abdellah, Laboratory Natural Resources and Environment, Faculty Polydisciplinary of Taza, Route of Oujda, Taza Station, Morocco*

*Corresponding author, Email : sibnahmed@yahoo.fr; arouyakhali@gmail.com

This study is part of the characterization of surface water chemistry and the determination of the origin of chemical elements present in the waters of the Taza region of Morocco. To carry out this investigation, 60 samples were collected in July 2017 and January 2018 (30 samples per month). The principal component analysis (PCA) applied to the data sets resulted in three significant factors accounting for 89.46% of the total variance. The F1 factor expresses 53.51% of the total information, called the factor of salinization, water pollution by nitrogen compounds and seasonality. Factors F2 and F3 can be called anthropogenic pollution and seasonality factors. In addition, the stations are well typed and therefore, well structured by their physico-chemical data. In addition, hierarchical cluster analysis (CA) grouped the twenty-five physico-chemical variables studied into five distinct groups related to water-rock interaction, agriculture and anthropogenic sources. Therefore, this work indicates that multivariate statistical methods are an excellent exploratory tool for interpreting a complex set of water quality data and for understanding spatial variations that are useful and effective for water quality management.

KEYWORDS

Multivariate analysis, Surface water, Physico-chemical quality, Northern Morocco

REFERENCES

1. Tampo, L., *et al.* 2018. Application of statistical methods to the hydrochemical study of the waters of a tropical hydrosystem: Case of the watershed of the Zio river (Togo). *European Sci. J.*, 11 (14): 204-225.
2. Haouchine, S. 2017. Research on faunal quality and ecology of macroinvertebrates benthic watercourses of Kabylia, mem. Magister BioSciences.
3. Foto, M.S., *et al.* 2011. Spatial evolution of the diversity of populations of benthic macro-invertebrates in an anthropized stream in a tropical environment (Cameroon). *European J. Sci. Res.*, 55(2): 291-300.
4. OMS. 2012. Guidelines for drinking water quality (4th edn). World Health Organization, Geneva, Switzerland. pp 307-447.
5. Iscen, C.F., *et al.* 2008. Application of multivariate statistical techniques in the assessment of surface water quality in Uluabat Lake, Turkey. *Env. Monitor. Assess.*, 144:269-276.
6. Mustapha, A. and A. Abdu. 2012. Application of principal component analysis and multiple regression models in surface water quality assessment. *J. Env. Earth Sci.*, 2(2):16-23.
7. Nourbakhsh, Z. and H. Yousefi. 2017. Presenting a conceptual model of data collection to manage the groundwater quality. *J. Water Land Develop.*, 35:149-160. DOI:10.1515/jwld-2017-0079.
8. Tiwari, R.N. 2011. Assessment of groundwater quality and pollution potential of Jawa block Rewa district, Madhya Pradesh. *Int. Academy Ecol. Env. Sci.*, 1(3-4):202-212.
9. El Morhit, M., *et al.* 2008. Impact of hydraulic development on the quality of water and sediment in the Loukkos estuary (Atlantic coast, Morocco). *Bulletin Sci. Institute, Rabat, Earth Sci.*, 30:39-47.
10. Belkhir, L., *et al.* 2010. Application of multivariate statistical methods and inverse geochemical modeling for characterization of groundwater: A case study of Ain Azel plain (Algeria). *Geoderma.*, 159: 390-398.
11. Makhoukh, M., *et al.* 2011. Contribution to the physico-chemical study of surface water from the Moulouya wadi (Eastern Morocco). *Larhyss J.*, 9: 149-169.

12. Lamrani, H., *et al.* 2014. Evaluation of the physico-chemical and bacteriological quality of the Boufekrane wadi in the vicinity of the effluents of the city of Meknes (Morocco). *ScienceLib Mersenne Editions*. 3:111-112.
13. Belghiti, M.L., *et al.* 2018. Physico-chemical characteristics of the water from certain wells used as a portable water source in rural areas in the region of Meknes (Morocco). *Larhyss J.*, 14:21-36.
14. Metrak, M., *et al.* 2014. Age and landuse as factors differentiating hydrochemistry and plant cover of astatic ponds in post-agricultural landscape. *J. Water Land Develop.*, 21:29-37. DOI: 10.2478/jwld-2014-0011.
15. Bak, L., *et al.* 2014. Preliminary assessment of silting and the quality of bottom sediments in a small water reservoir. *J. Water Land Develop.*, 21: 47-53. DOI: 10.2578/jwld-2014-0013.
16. Soro, T.D. 2017. Evolution of water resources in the Haut Bandama watershed in Tortiya (North ern Cote d'Ivoire) in a context of climate variability and change: Hydrological, hydrogeological and hydrochemical impact. PhD Thesis. Felix University, Houphouet-Boigny, Ivory Coast.
17. Eblin, S.G., *et al.* 2016. Hydrochemistry of surface water in the Adiake region (south-eastern coast of Cote d'Ivoire). *J. Appl. Biosci.*, 75:6259-6271.
18. Reggam, A., *et al.* 2015. Physico-chemical quality of the waters of the Oued Seybouse (Northeastern Algeria): Characterization and principal component analysis. *J. Mater. Env. Sci.*, 6(5): 1417-1425.
19. Al Ahmadi, M.E. 2015. Multivariate statistical analysis of groundwater quality in Wadi Ranyah, Saudi Arabia JAKU. *Earth Sci.*, 21(2): 29-46.
20. Hachol, J., *et al.* 2017. Applying the analytical hierarchy process (AHP) into the effects assessment of river training works. *J. Water Land Develop.*, 35: 63-72. DOI: 10.1515/jwld-2017-0069.
21. Boussaha, S., *et al.* 2017. Wadi Bounamoussa's waters quality in the north-east of Algeria: Statistical treatment of some physical and chemical parameters. *J. Water Land Develop.*, 34:77-83. DOI: 10.1515/jwld-2016-0040.
22. Sayad, L., *et al.* 2017. Hydrochemical study of Dreaan-Annaba aquifer system (NE Algeria). *J. Water Land Develop.*, 34(7-9):259-263. DOI: 10.1515/jwld-2017-0061.
23. Lakhili, F., *et al.* 2015. Study of the physico-chemical quality and metallic contamination of surface water in the Beht watershed (Morocco). *European Sci. J.*, 11 (11).
24. Corneille, B., *et al.* 2017. Physico-chemical characterization of borehole water in the villages of Tanlili and Lilgomde in the northern region of Burkina Faso - Correlation between the physico-chemical parameters. *Africa Sci.*, 13(6): 325-337.
25. Diallo, A.D., *et al.* 2016. The value of statistical analysis methods in the management of monitoring the physico-chemical quality of water on the right bank of the Senegal river. *Larhyss J.*, 17: 101-114.
26. Ahoussi, K.E., *et al.* 2017. Hydrochemical and microbiological study of spring water from the mountainous west of Cote d'Ivoire: Case of the village of Mangouin-Yrongouin (sub-prefecture of Biankouman). *J. Appl. Biosci.*, 63: 4703-4719.
27. Buhungu, S., *et al.* 2018. Spatio-temporal characterization of the water quality of the Kinyankonge river, tributary of Lake Tanganyika, Burundi. *Int. J. Biol. Chem. Sci.*, 12(1): 576-595.
28. Mezbour, R., *et al.* 2018. Evaluation of organic pollution index and the bacteriological quality of the water of the Lake of birds (EL Tarf East-Algerian). *J. Mater. Env. Sci.*, 9(3): 971-979. DOI: 10.26872/jmes.2018.9.3.108.
29. Taous, A., *et al.* 2004. Impacts of lateral cones on hydrodynamic and hydrological functioning. Current morphosedimentary morphology of prerifaine valley bottoms: Case of Oued Larbaa (Morocco). *J. Alpine Geography*. 92(92-1): 17-28.
30. Rodier, J., *et al.* 2009. The water analysis (9th edn). DUNOD, Paris. pp 1203.
31. Rouabhia, A.E.K. and L. Djabri. 2010. Irrigation and the risk of salt pollution. Example of groundwater from the Miocene aquifer of the El Ma Labiod plain. *Larhyss J.*, 8:55-67.
32. Laaksoharju, M., *et al.* 1999. Multivariate mixing and mass balance (M3) calculations, a new tool for decoding hydrogeochemical information. *Appl. Geochem.*, 14: 861-871.
33. Tauler, R., *et al.* 2000. Multivariate correlation between concentrations of selected herbicides and derivatives in outflows from selected US mid-western reservoirs. *Env. Sci. Tech.*, 34(16): 3307-3314.
34. Singh, K.P., *et al.* 2005. Chemometric data analysis of pollutants in wastewater - A case study. *Anal. Chim. Acta*. 532(1): 15-25.
35. Varol, M., *et al.* 2012. Spatial and temporal variations in surface water quality of the dam reservoirs in the Tigris river basin, Turkey. *Catena*. 92: 11-21.
36. Ouyang, Y. 2005. Evaluation of river water quality monitoring stations by principal component analysis. *Water Res.*, 39(12): 2621-2635.
37. Pardo., R., *et al.* 2004. Application of two- and three-way principal component analysis to the interpretation of chemical fractionation results obtained by the use of the BCR procedure. *Anal. Chim. Acta*. 523(1): 125-132.
38. Dagnelie., P. 2018. Theoretical and applied statistics: Inferences to one and two-dimensional. Bruxelles University, Boeck and Larcier. pp 736.

39. Lattin, J.M., J. D. Carroll and P. E. Green. 2003. Analyzing multivariate data. Duxbury applied series. Thomson Brooks/Cole, Pacific Grove, USA.
40. McKenna, J. 2003. An enhanced cluster analysis program with bootstrap significance testing for ecological community analysis. *Env. Modelling Software*. 18: 205-220.
41. Otto, M. 1998. Multivariate methods. In Analytical chemistry. Ed R. Kellner, J. M. Mermet, M. Otto and H. M. Widmer. Wiley-VCH, Weinheim.
42. Foucart, T. 1983. A new approach to the STATIS method. *Review Appl. Statistics*. 31: 61-75.
43. Dragon, K. 2012. Application of factor analysis to study contamination of a semi-confined aquifer (Wielkopolska Buried Valley aquifer, Poland). *J. Hydrol.*, 331(1): 272-279.
44. Attoui, B. *et al.* 2018. Assessment of groundwater vulnerability to pollution using the Kherici's method in the Talezza plain Collo region (NE Algeria). *J. Water Land Develop.*, 33: 23-30. DOI: 10.1515/jwld-2017-0015.
45. Obiefuna, G.I. and A. Sheriff. 2017. Assessment of shallow groundwater quality of Pindiga Gombe area, Yola area, NE, Nigeria for irrigation and domestic purposes. *Res. J. Env. Earth Sci.*, 3(2): 131-141.
46. Robinove, C.J., *et al.* 1958. Saline water resource of North Dakota (Water supply paper 1428). U.S. Government Printing Office, Washington, D.C. DOI: 10.3133/wsp1428.
47. WHO. 2011. Wastewater use in agriculture and aquaculture: Recommendation health advisories. World Health Organization, Geneva. pp 82.
48. Abbou, M.B., *et al.* 2014. Degradation of water quality in the alluvial aquifer of Wadi Larbaa by wastefrom the city of Taza (Morocco). *Int. J. Innovation Sci. Res.*, 10(2): 282-294.
49. Naoura, J., *et al.* 2015. Assessment of water quality of the Inaouene river, Northern Morocco. *Int. J. Innovation Appl. Studies*. 10(1): 60-66.
50. Hebert, S. 2017. Monitoring the water quality of rivers and small streams. Department for Monitoring State of the Environment, Ministry of the Environment, Government of Quebec.
51. N.M. 2002. Moroccan water quality standard. Joint Order of the Minister of Equipment and of the Minister responsible for land use planning, urbanism, housing and the environment. Bull. Off. no. 5062. 30 Ramadan 1423.
52. Korfali., S.I. and M. Jurdi. 2017. Suitability of surface water for domestic water use: Awali river case study. *European Water*. 35: 3-12.
53. Mmualefhe, L.C. and N. Torto. 2011. Water quality in the Okavango Delta. *Water SA*. 37: 411-418.
54. Pitt, R., *et al.* 2004. Review of historical street dust and dirt accumulation and washoff data. In Effective modeling of urban systems, monograph 13. Ed W. James *et al.* CHI, Guelph, Ontario.
55. Wong, T.H.F. 2006. Australian runoff quality - A guide to water sensitive urban design. Engineers Australia.

Identification And Control Of Giant African Snail In Cauliflower Farms By Low Frequency Sound Measurement Technique And Back Propagation Neural Network

Adeline Sneha J.¹, Rekha Chakravarthi^{2*}, Nor Hisham Khamis³ and Joshua Amarnath D.⁴

1. Sathyabama Institute of Science and Technology, Chennai - 600 119, Tamil Nadu, India

2. Sathyabama Institute of Science and Technology, Department of Electronics and Telecommunication Engineering, Chennai - 600 119, Tamil Nadu, India

3. Universiti Teknologi Malaysia, Department of Communication Engineering, Malaysia

4. Sathyabama Institute of Science and Technology, Department of Chemical Engineering, Chennai - 600 119, Tamil Nadu, India

*Corresponding author, Email : rekha_2705@yahoo.co.in; j.adelinesneha@gmail.com

The Giant African snail is the world's largest, destructive and most damaging land pest in agricultural plantations. These pests are sometimes called as the natural enemies of farmers. High devouring capacity of these snails causes heavy damages to vegetation. In this work, various snails are collected and placed in the room temperature. The snails were fed with the consumables, such as cauliflower, tomatoes and soil, indeed the sound frequency is measured. In addition, the various activities also monitored and the respective frequencies are observed. For instance, the movement of a snail, retraction of the head in shell, head movements are recorded from the acoustic room with the recorder. However, monitoring the snail low level sound were challenging for recording. Thus the high precision device is required for recording the signals. The recording has been done for a number of days in order to evade uncertainty. The recorded accurate signals are chosen for the analysis. These signals are analyzed in the time and frequency domain. The statistical features are extracted from the signal. Further, the power spectral density, sound pressure level and equal loudness contour of the signals are calculated. The wide range of frequency of the snail of its various movements is determined. The extracted statistical features from the signals are fed into the neural network for training. Back propagation algorithm is used for classification along with the snail's signal, few noise signals also fed into the neural network to identify the effectiveness of network in the identification of the pest. It gives 99% efficiency in identifying the snail's signal. This work is also compared with capacitance based pest identification. This work has been experimentally validated with low-cost acoustic sensor that wirelessly communicates to the base, to monitor and control snail activity, the work suggested in this paper is a non-destructive form of pest identification which does not harm the crops as well as the environment.

KEYWORDS

Pest identification, African snail, Acoustic, Crops protection

REFERENCES

1. Oerke, E.C. 2006. Crop losses to pests. *J. Agric. Sci.*, 44(1): 31-43.
2. IUCN SSG. 2012. Global invasive species database. Invasive Species Specialist Group.
3. Nelson, S. 2012. The Giant African snail. Hanai' Ai/The food provider.
4. Sridhar, V., et al. 2013. Severe occurrence of the giant African snail, *Achatina fulica* (Bowdich) (Stylommatophora: Achatinidae) in Kolar district, Karnataka. *Pest Manage. Horticultural Ecosystems*. 18(2): 228-230.
5. Nelson, S. 2012. Injuries caused by the giant African snail to papaya. College of Tropical Agriculture and Human Resources, University of Hawai'i, Manoa, United States.
6. Raut, S. and G. Barker. 2002. *Achatina fulica* Bowdich and other achatinidae as pests in tropical agriculture. In Molluscs as crop pests. Ed G.M. Barker. CAB International, Wallingford. pp 55-114.
7. Roda, A., et al. 2018. A trapping strategy to aid giant African snail (*Lissachatina fulica*) eradication programmes. *PLoS one*. 13(9).

8. Bagarinao, T. and I. Lantin-Olaguer. 2000. From triphenyltins to integrated management of the pest snail *Cerithidea cingulata* in mangrove-derived milkfish ponds in the Philippines. *Hydrobiologia*. 437(1-3):1-6.
9. Kulsantiwong, J., et al. 2013. DNA barcode identification of freshwater snails in the family Bithyniidae from Thailand. *PloS one*. 8(11).
10. Ferro, E., et al. 2014. Wireless sensor mote for snail pest detection. *Sensors IEEE*. DOI: 10.1109/ICSENS.2014.6984946.
11. Garcia-Lesta, D., et al. 2016. Live demonstration: Wireless sensor network for snail pest detection. Proceedings of IEEE International Symposium on Circuits and systems. pp 2371.
12. Garcia-Lesta, D., et al. 2015. Capacitance-based wireless sensor mote for snail pest detection. Proceedings of IEEE Sensors applications Symposium. pp 1-6.
13. Dunstan, D.J. and D.J. Hodgson. 2014. Snails home. *Physica Scripta*. 89(6).
14. Wu, J.Y., et al. 2014. Identification of optimum scopes of environmental factors for snails using spatial analysis techniques in Dongting lake region, China. *Parasites Vectors*. 7. DOI: 10.1186/1756-3305-7-216.
15. Kristanto, A.H., et al. 2019. Survey on egg and fry production of giant gourami (*Osphronemus goramy*): Current rearing practices and recommendations for future research. *J. World Aquaculture Society*. 51(1).
16. Emelue, G.U. and E.A. Omonzogbe. 2018. Growth performance of African giant land snails (*Archachatina marginata*) fed with feed formulated with different calcium sources. *Malaysian J. Sustainable Agric.*, 2(1): 1-4.
17. Hausdorf, B. 2018. The giant African snail *Lissachatina fulica* as potential index fossil for the Anthropocene. *Anthropocene*. 23:1-4.

Investigation Of Groundwater Quality For Agricultural Use In A Lateritic Soil Belt

Mobarak Hossain and Pulak Kumar Patra*

Visva-Bharati, Department of Environmental Studies, Institute of Science, Santiniketan - 731 235, Birbhum

*Corresponding author, Email : pulakpatra@visva-bharati.ac.in

Groundwater is a vulnerable natural resource. The problems of its quality for its designated purposes, such as drinking, domestic, industrial and agricultural uses are gaining attention, especially in developing countries. The present study was undertaken to investigate the suitability of groundwater for irrigational purpose in Bolpur block of Birbhum district of West Bengal. Major cations in groundwater samples were found in descending order as $\text{Ca}^{2+} > \text{Na}^+ > \text{Mg}^{2+} > \text{K}^+$ whereas, anions were $\text{HCO}_3^- > \text{Cl}^- > \text{SO}_4^{2-} > \text{NO}_3^-$. Water quality for irrigation was assessed through water quality parameters, such as electrical conductivity (EC), sodium absorption ratio (SAR), soluble sodium percentage (SSP), residual sodium carbonate (RSC), Kelly's ratio (KR) and permeability index (PI). The values of these water quality parameters were found to be suitable for irrigation. Major hydrochemical facies of groundwater of the study area were normal earth alkaline water with prevailing bicarbonate, that is $\text{Ca}^{2+}\text{-Mg}^{2+}\text{-HCO}_3^-$ type. Local geological setting and rock-water interactions are a principal factor that governed the formation of groundwater chemistry of the study area. The outcome of the present study may be helpful to the farmers, researchers and policymakers to formulate an effective groundwater management plan.

KEYWORDS

Irrigational water quality indices, Hydrochemistry, Piper diagram, Gibb's diagram, Bolpur block

REFERENCES

1. Srivastava, S.K. 2019. Assessment of groundwater quality for the suitability of irrigation and its impacts on crop yields in the Guna district, India. *Agric. Water Manage.*, 216:224-241. DOI: 10.1016/j.agwat.2019.02.005.
2. Gomez, V.M.R., et al. 2017. Groundwater quality impacted by landuse / land cover change in a semi-arid region of Mexico. *Groundwater Sustain. Develop.*, 5:160-167. DOI: 10.1016/j.gsd.2017.06.003.
3. Pazand, K., et al. 2018. Identification of the hydrogeochemical processes and assessment of groundwater in a semi-arid region using major ion chemistry: A case study of Ardestan basin in Central Iran. *Groundwater Sustain. Develop.*, 6:245–254. DOI: 10.1016/j.gsd.2018.01.008.
4. Todd, D.K. and L.W. Mays. 2005. Groundwater hydrology (3rd edn). John Wiley & Sons, United States. pp 636.
5. Alemu, M.M. and F.Y. Desta. 2017. Irrigation water quality of river Kulfo and its implication in irrigated agriculture, southwest Ethiopia. *Int. J. Water Resour. Env. Eng.*, 9:127-132.
6. Nishanthiny, S.C., et al. 2010. Irrigation water quality based on hydrochemical analysis, Jaffna, Sri Lanka. *American Eurasian J. Agric. Env. Sci.*, 7:100-102.
7. Zafar, S., et al. 2017. Assessing impact of effluent discharge on irrigation water quality in southern region of Khyber Pakhtunkhwa, Pakistan. *Env. Monit. Assess.*, 189(4):156.
8. Singh, P.K., et al. 2018. Evaluation of the surface water quality index of Jharia coal mining region and its management of surface water resources. *Env. Poll.*, 429-437.
9. Barik, R. and S.K. Pattanayak. 2019. Assessment of groundwater quality for irrigation of green spaces in the Rourkela city of Odisha. *Groundwater Sustain. Develop.*, 8:428-438. DOI: 10.1016/j.gsd.2019.01.005.
10. Iqbal, J., et al. 2018. Hydrochemical processes determining the groundwater quality for irrigation use in an arid environment: The case of Liwa aquifer, Abu Dhabi, United Arab Emirates. *Groundwater Sustain. Develop.*, DOI: 10.1016/j.gsd.2018.06.004.
11. Khanoranga and S. Khalid. 2018. An assessment of groundwater quality for irrigation and drinking purposes around brick kilns in three districts of Balochistan province, Pakistan, through water quality index and multivariate statistical approaches. *J. Geochem. Exploration*. DOI: 10.1016/j.gexplo. 2018.11.007.

12. Brindha, K. and L. Elango. 2011. Hydrochemical characteristics of groundwater for domestic and irrigation purposes in Madhuranthakam, Tamil Nadu. *Earth Sci. Res. J.*, 15(2):101-108.
13. Kalpana, L. and L. Elango. 2013. Assessment of groundwater quality for drinking and irrigation purposes in Pambar river sub-basin, Tamil Nadu. *Indian J. Env. Prot.*, 33(1):1-8.
14. Nag, S.K. and S. Das. 2014. Groundwater quality for irrigation and domestic purposes – A GIS based case study of Suri I and II blocks, Birbhum district, West Bengal. *Int. J. Adv. Earth Env. Sci.*, 2(1):25-38.
15. Nag, S.K. and B. Suchetana. 2016. Groundwater quality and its suitability for irrigation and domestic purposes: A study in Rajnagar block, Birbhum district, West Bengal. *J. Earth Sci. Climate Change*. 7(2):337. DOI: 10.4172 /2157-7617.1000337.
16. APHA. 2012. Standard methods for the examination of water and wastewater (22nd edn). American Public Health Association, Washington D.C., United States.
17. WHO. 2011. Guidelines for drinking-water quality (4th edn). World Health Organization, Geneva.
18. Richards, L.A. 1954. Diagnosis and improvement of saline and alkali soils. Agricultural handbook 60. United States Department of Agriculture.
19. Todd, D.K. 1980. Groundwater hydrology (2nd edn). John Wiley & Sons Inc., USA.
20. Raghunath, H.M. 1987. Groundwater (2nd edn). Wiley Eastern Pvt. Ltd., New Delhi.
21. Eton, F.M. 1950. Significance of carbonate in irrigation waters. *Soil Sci. J.*, 67(3):128-133.
22. Abdulhusein, F.M. 2018. Hydrochemical assessment of groundwater of Dibdibba aquifer in Al-Zubair area, Basra, south of Iraq and its suitability for irrigation purposes. *Iraqi J. Sci.*, 59(1A):135-143.
23. Ravikumar, P., et al. 2011. Geochemistry of groundwater and groundwater prospects evaluation, Anekal taluk, Bangalore urban district, Karnataka. *Env. Monit. Assess.*, 179: 93-112.
24. Kelly, W.P. 1940. Permissible composition and concentration of irrigated waters. *Proceedings of the American Society of Civil Engineers*. 66:607-613.
25. Doneen, L.D. 1975. Water quality for irrigated agriculture. In *Plants in saline environments*. Ed A. Poljakoff-Mayber, et al. Springer-Verlag Berlin Heidelberg. pp 56-76.
26. Alharbi, T.G. 2018. Identification of hydrogeo-chemical processes and their influence on groundwater quality for drinking and agricultural usage in Wadi Nisah, Central Saudi Arabia. *Arabian J. Geosci*. DOI: 10.1007/s12517-018-3679-z.
27. Naseem, S., S. Hamza and E. Bashir. 2010. Groundwater geochemistry of Winder agricultural farms, Balochistan, Pakistan and assessment for irrigation water quality. *European Water*. 31:21-32.
28. Zaidi, F.K., et al. 2015. Reverse ion exchange as a major process controlling the groundwater chemistry in an arid environment: A case study from northwestern Saudi Arabia. *Env. Monit. Assess.*, 187(10):1-18.
29. Awais, M., et al. 2017. Evaluating groundwater quality for irrigated agriculture: Spatio-temporal investigations using GIS and geostatistics in Punjab, Pakistan. *Arab J. Geosci.*, 10(23):510.
30. Chitsazan, M., et al. 2017. Hydrochemical characteristics and quality assessment of urban groundwater in Urmia city, NW Iran. *Water Supply*. 17 (5): 1410-1425
31. Hussain, Y., et al. 2017. Modelling the vulnerability of groundwater to contamination in an unconfined alluvial aquifer in Pakistan. *Env. Earth Sci.*, 76(2):84.
32. Obiefuna, G.I. and A. Sheriff. 2011. Assessment of shallow groundwater quality of Pindiga Gombe area, Yola area, NE, Nigeria for irrigation and domestic purposes. *Res. J. Env. Earth Sci.*, 3(2):131-141.
33. Piper, A.M. 1944. A graphic procedure in the geochemical interpretation of water analyses. *Trans. American Geophy. Union*. 25:914-923.
34. Gibbs, R.J. 1970. Mechanisms controlling world water chemistry. *Sci.*, 170:1088-1090.

Closed Pattern Mining And Causal Analysis Of Pollution Data

S. Sharmiladevi and S. Siva Sathya*

Pondicherry University, Department of Computer Science, Puducherry - 605 014, India

*Corresponding author, Email : ssivasathya@gmail.com; sharmiladevi94@gmail.com

Mining sequential patterns are of great importance in recent years, as it unveils some of the unknown associative relationships between observations. While in mining sequential patterns many intermediate sequences have to be generated, which is a computationally challenging task when compared to frequent patterns of mining. CloFAST is an algorithm which mines closed sequences without candidate maintenance. Also, CloFAST requires only one step to check closure and prune the search space. It can mine long closed sequences effortlessly from large datasets. In this work, a closed sequential pattern mining of $PM_{2.5}$ pollutant in Delhi is done using CloFAST. Delhi, the capital of the second most populous country on earth has been suffering from severe air pollution problem. Delhi is getting polluted due to diverse reasons, like its geography, burning crop stubble in neighbouring states, vehicular emission, etc. Some of the critical air pollutants found in Delhi are PM_{10} , $PM_{2.5}$, nitrogen oxide, sulphur oxide, carbon monoxide, ozone. The main pollutant being particulate matter ($PM_{2.5}$) as it causes serious health problems when it enters into the alveoli of human lungs. Various micro-level analysis of air pollution is being carried out recently. But macro-level analysis is also required in order to obtain a clear understanding on a broader scale. The patterns obtained are given as knowledge for causal analysis done using the FCI algorithm.

KEYWORDS

Air pollution, $PM_{2.5}$ pollutant, Closed sequential mining, Particulate matter, Delhi

REFERENCES

1. Guttikunda, S.K. and B.R. Gurjar. 2012. Role of meteorology in seasonality of air pollution in megacity Delhi. *Env. Monitor. Assess.*, 184(5): 3199-3211. DOI:0.1007/s0661-011-2182-8.
2. Zhao, C. and G. Song. 2017. Application of data mining to the analysis of meteorological data for air quality prediction: A case study in Shenyang. *IOP Conference Series: Earth Env. Sci.*, 81. DOI: 10.1088/1755-1315/81/1/012097.
3. Nagpure, A.S., B.R. Gurjar and J. Martel. 2014. Human health risks in national capital territory of Delhi due to air pollution. *Atmos. Poll. Res.*, 5(3): 371-380.
4. Ming, L., et al. 2017. $PM_{2.5}$ in the Yangtze river delta, China: Chemical compositions, seasonal variations and regional pollution events. *Env. Poll.*, 223:200-212. DOI:10.1016/j.envpol.2017.01.013.
5. Times of India. 2018. Usual suspects: Vehicles, industrial emissions behind foul play all year. Available: <https://timesofindia.indiatimes.com/city/delhi/usual-suspects-vehic-les-industrial-emissions-behind-foul-play-all-year/articleshow/66228517.cms>.
6. Fournier-Viger, P., et al. 2017. A survey of sequential pattern mining. *Data Sci. Pattern Recognition*. 1(1): 54-77.
7. Wang, L., et al. 2018. Effective lossless condensed representation and discovery of spatial co-location patterns. *Information Sci.*, 436-437: 197-213.
8. Fumarola, F., et al. 2016. CloFAST: Closed sequential pattern mining using sparse and vertical id-lists. *Knowledge Information Systems*. 48(2): 429-463. DOI:10.1007/s 10115-015-0884-x.
9. Srikant, R. and R. Agrawal. 1996. Mining sequential patterns: Generalizations and performance improvements. In *Advances in database technology (vol 1057)*. Ed P. Apers, M. Bouzeghoub and G. Gardarin. Springer-Verlag Berlin Heidelberg.
10. Zaki, M.J. 2001. SPADE: An efficient algorithm for mining frequent sequences. *Machine Learning*. 42(1-2): 30.
11. Fournier-Viger, P., et al. 2014. Fast vertical mining of sequential patterns using co-occurrence information. In *Advances in knowledge discovery and data mining (vol 8443)*. Springer International Publishing. pp 40-52.
12. Ayres, J., et al. 2002. Sequential pattern mining using a bitmap representation. KDD '02: Proceedings of the eighth ACM SIGKDD International Conference on Knowledge discovery and data mining. pp 429-435.

13. Yang, Z., Y. Wang and M. Kitsuregawa. 2007. LAPIN: Effective sequential pattern mining algorithms by last position induction for dense databases. In *Advances in databases: Concepts, systems and applications* (vol 4443). Springer-Verlag Berlin Heidelberg. pp 1020-1023.
14. Yan, X., J. Han and R. Afshar. 2003. CloSpan: mining: Closed sequential patterns in large data sets. *Proceedings of the third SIAM International Conference on Data Mining*. DOI:10.1137/1.9781611972733.15.
15. Wang, J. and J. Han. BIDE: Efficient mining of frequent closed sequences. *Proceedings of the 20th International Conference on Data engineering*. pp 79-90. DOI:10.1109/ICDE. 2004.1319986.
16. Spirtes, P., C. Glymour and R. Scheines. 2000. *Causation, prediction and search* (2nd edn). MIT Press.
17. Jiang, L. and L. Bai. 2018. Spatio-temporal characteristics of urban air pollutions and their causal relationships: Evidence from Beijing and its neighbouring cities. *Scientific Reports*. 8(1). DOI: 10.1038/s41598-017-18107-1.
18. Li, X., *et al.* 2017. Discovering pollution sources and propagation patterns in urban area. *Proceedings of the 23rd ACM SIGKDD International Conference on Knowledge discovery and data mining*. pp 1863-1872. DOI: 10.1145/3097983.3098090.
19. Zhu, J.Y., C. Sun and V.O. K. Li. 2015. Granger-causality-based air quality estimation with spatio-temporal (S-T) heterogeneous big data. *Proceedings of the IEEE Conference on Computer communications workshops (INFOCOM WKSHPS)*, Hong Kong. pp 612-617. DOI:10.1109/INFOCOMW.2015. 7179453.
20. Zhu, J.Y., *et al.* 2016. A Gaussian bayesian model to identify spatio-temporal causalities for air pollution based on urban big data. *Proceedings of the IEEE Conference on Computer communications workshops (INFOCOM WKSHPS)*, San Francisco, USA. pp 3-8. DOI: 10.1109/INFOCOMW.2016.756 2036.
21. Yuan, Q., *et al.* 2014. Temporal variations, acidity and transport patterns of PM_{2.5} ionic components at a background site in the Yellow river delta, China. *Air Quality Atmos Health*. 7(2): 143-153.
22. Joshi, D., A.S. Sabitha and S. Sharma. 2016. Air pollution data analysis using time series clustering for IOT. *Int. J. Cont. Theory Applications*. 9(46): 12.
23. Sathya, D., J. Anu and M. Divyadharshini. 2017. Air pollution analysis using clustering algorithms. *Proceedings of the International Conference on Emerging trends in engineering, science and sustainable technology*. pp 4.
24. Zhang, H., Z. Wang and W. Zhang. 2016. Exploring spatiotemporal patterns of PM_{2.5} in China based on ground-level observations for 190 cities. *Env. Poll.*, 216: 559-567. DOI: 10.1016/j.envpol.2016.009.
25. Zhou, M., *et al.* 2016. Spatial and temporal patterns of air quality in the three economic zones of China. *J. Maps*. 12 (sup1): 156-162. DOI:10.1080/1744564 7.2016.1187095.
26. Dadhich, A.P., R. Goyal and P.N. Dadhich. 2018. Assessment of spatio-temporal variations in air quality of Jaipur city, Rajasthan. *The Egyptian J. Remote Sensing Space Sci.*, 21(2): 173-181. DOI: 10.1016/j.ejrs.2017.04.002.
27. Lin, G., *et. al.* 2013. Spatio-temporal variation of PM_{2.5} concentrations and their relationship with geographic and socio-economic factors in China. *Int. J. Env. Res. Public Health*. 11(1): 173-186.
28. Sharma, N., *et. al.* 2018. Forecasting air pollution load in Delhi using data analysis tools. *Procedia Computer Sci.*, 132: 1077-1085. DOI:10.1016/j. procs.2018.05.023.
29. Yang, G., J. Huang and X. Li. 2018. Mining sequential patterns of PM_{2.5} pollution in three zones in China. *J. Cleaner Production*. 170:388-398. DOI: 1016/j.jclepro.2017.09.162.
30. Bellinger, C., *et al.* 2017. A systematic review of data mining and machine learning for air pollution epidemiology. *BMC Public Health*. 17(1). DOI:10. 1186/s12889-017-4914-3.

Effect Of Mutation On Bacillus Species For The Degradation Of Azo Dye And Its Molecular Characterization Using AFLP

Himanshi Mangla¹, Hardik Pathak^{1*}, Saurabh Dave², Mahesh³ and D. P. Jaroli⁴

1. JECRC University, Department of Biotechnology, Jaipur - 303 905, Rajasthan, India

2. JECRC University, Department of Chemistry, Jaipur - 303 905, Rajasthan, India

3. Azyme Biosciences, Bangalore - 560 069, Karnataka, India

4. Board of Secondary Education Rajasthan, Ajmer - 305 001, Rajasthan, India

*Corresponding author, Email : hardikaeshu@gmail.com; himanshimangla30@gmail.com

Azo dyes are frequently used synthetic dye in the textile industries. It is a xenobiotic compound due to the presence of the sulphonic group. These azo dyes release through dye effluent or industrial waste resulting in an acute effect on the environment and also on human health. The present investigation was an attempt to know the degradation of Para Red azo dye at physico-chemical parameters, such as pH, temperature and concentration. An experiment was carried out to degrade the Para Red azo dye using bacterial isolates from the dye contaminated soil. From all the isolates, only Bacillus species was able to degrade the dye. As it has the ability to produce the azo reductase enzyme to catalyze the Para Red azo dye. Approximately 53% of degradation was observed after optimization. But after mutations, the enzyme activity was enhanced and 68.2% degradation was observed. After purification, an increase in fold purification and a gradual decrease in the yield of about 1.13% and 70%, respectively was observed.

KEYWORDS

Azo dye, Effluents, Azo reductase, Enzyme, Mutations

REFERENCES

1. Lacasse, K. and W. Baumann. 2007. Textile chemicals: Environmental data and facts. Springer-Verlag Berlin Heidelberg.
2. McMullan, G., et al. 2001. Microbial decolourization and degradation of textile dyes. *Appl. Microbiol. Biotech.*, 56(1-2): 81-87.
3. Rajendra, R., A.M. Hasab and S.K. Sundaram. 2012. Development of microbial consortium for the biodegradation and biodecolourization of textile effluents. *J. Urban Env. Eng.*, 6: 36-41.
4. Dawkar, V.V., et al. 2008. Biodegradation of disperse textile dye brown 3 REL by newly isolated Bacillus sp. VUS. *J. Appl. Microbiol.*, 105(1):14-24.
5. Gandhi, S.A., et al. 2015. Skin cancer epidemiology, detection and management. *Medical Clinics North America*. 99(6): 1323-1335.
6. Brown, M.A. and S.C. De Vito. 1993. Predicting azo dye toxicity. *Critical Reviews Env. Sci. Tech.*, 23(3): 249-324.
7. So, C.M., et al. 2002. Degradation of azo dye procion red MX-5B by photocatalytic oxidation. *Chemosphere*. 46(6): 905-912.
8. Waghmode, T.R., et al. 2019. Sequential photocatalysis and biological treatment for the enhanced degradation of the persistent azo Methyl Red. *J. Hazard. Mater.*, 371: 115-122.
9. Verma, P. and D. Madamwar. 2003. Decolourization of synthetic dyes by a newly isolated strain of *Serratia marcescens*. *World J. Microbiol. Biotech.*, 19(6): 615-618.

Sorption Of Cu(II) Ions On Orange Peel

Shubhangi S. Malkhede^{1*} and Y. R. M. Rao²

1. MIT, Department of Civil Engineering, Kothrud, Pune - 411 038, Maharashtra, India

2. Dr. Pauls Engineering College, Pulichapallam, Villupuram - 605 109, Tamil Nadu, India

*Corresponding author, Email : shub_shekokar13@rediffmail.com ; dryrmrao@rediffmail.com

An adsorption process is the most effective technique and used for the elimination or reduction of metal ions from wastewater. However, the process is not economical since the activated carbon used as a sorbent is costlier. Hence investigators are inspecting to identify cost-effective, efficient, locally available adsorbent materials. In this research work, attempts have been made to check the efficiency of powdered orange peel (POP) with powdered activated carbon (PAC) in removing Cu²⁺ ions from the aqueous solution. Experiments have been carried out using the batch mode method. In this removal process, the influence of various physico-chemical parameters, like pH, contact time, adsorbent dose, Cu²⁺ ions awareness and adsorbent particle size at the adsorption of Cu²⁺ ions through each POP and PAC were investigated. Results found out that the 69.75% elimination of Cu²⁺ ions was determined with powdered orange peel at pH 5, contact time 60 min, adsorbent dose 3.5 g/L and Cu²⁺ ions concentration of 200 ppm. Whereas, the powdered activated carbon has shown elimination of 77.4% at pH 5, contact time of 45 min, adsorbent dose 2.0 g/L and Cu²⁺ ions concentration of 500 ppm. Equilibrium records had been analyzed using the Langmuir, Freundlich model.

KEYWORDS

Activated carbon, Adsorbent, Adsorption, Batch mode technique, Contact time, Heavy metals, Isotherm, Powdered orange peel

REFERENCES

1. Dundar M., C. Nuhoglu and Y. Nuhoglu. 2008. Biosorption of Cu(II) ions onto the litter of natural trembling poplar forest. *J. Hazard. Mater.*, 151(1): 86-95.
2. Verma, R. and P. Dwivedi. 2013. Heavy metal water pollution- A case study. *Recent Res. Sci. Tech.*, 5(5): 98-99.
3. Krishna, R.H. and A.V.V.S. Swamy. 2012. Physico-chemical key parameters, Langmuir and Freundlich isotherm and Lagergren rate constant studies on the removal of divalent nickel from the aqueous solutions onto powder of calcined brick. *Int. J. Eng. Res. Develop.*, 4(1): 29-38.
4. Ghasemi, M., et al. 2014. Kinetic and equilibrium study of Ni(II) sorption from aqueous solutions onto *Peganum harmala* L. *Int. J. Env. Sci. Tech.*, 11: 1835-1844.
5. Bhargava, A. and A.K. Minocha. 2006. Conventional and non-conventional adsorbents for removal of pollutants from water – A review. *Indian J. Chem. Tech.*, 13: 203-217.
6. Okoli, J. and I. Ezuma. 2014. Studies of heavy metals by low-cost adsorbents. *J. Appl. Sci. Env. Manage.*, 18(3): 443-448.
7. Boumchita, S., et al. 2017. Application of peanut shell as a low-cost adsorbent for the removal of anionic dye from aqueous solutions. *J. Mater. Env. Sci.*, 8(7): 2353-2364.
8. Paramasivam, M. and C. Namasivayam. 1995. Removal of Ni(II) from aqueous solution and nickel plating industry wastewater using agricultural waste: Peanut hull. *Waste Manage.*, 15(1): 63-68.
9. Geetha, K.S. and S.L. Belagali. 2013. Removal of heavy metals and dyes using low-cost adsorbents from aqueous medium- A review. *IOSR J. Env. Sci. Toxicol. Food Tech.*, 4(3):56-68.
10. Altundogan, H.S., N.E. Arslan and F. Tumen. 2007. Copper removal from aqueous solutions by sugar beet pulp treated by NaOH and citric acid. *J. Hazard. Mater.*, 149(2): 432-439.
11. Benaissa, H. and M.A. Elouchdi. 2007. Removal of copper ions from aqueous solutions by dried sunflower leaves. *Chem. Eng. Processing*. 46(7): 614-622.
12. Khorraei, M., et al. 2007. Copper biosorption from aqueous solutions by sour orange residue. *J. Hazard. Mater.*, 149(2): 269-274.

13. Saeed, A., M.W. Akhter and M. Iqbal. 2005. Removal and recovery of heavy metals from aqueous solution using papaya wood as a new biosorbent. *Separation Purification Tech.*, 45(1): 25-31.
14. Rahman, F.B.A., M. Akter and M. Z. Abedin. 2013. Dyes removal from textile wastewater using orange peels. *Int. J. Sci. Tech. Res.*, 2(9):47-50.
15. Sha, L. 2009. Adsorption of Cu^{2+} and Cd^{2+} from aqueous solution by mercapto-acetic acid modified orange peel. *Colloids Surfaces B Biointerfaces*. 73:1-4.
16. Rao, V.V.B., C. Sailu and D.K. Sandilya. 2007. An experimental study of liquid-particle flow in circulating fluidized bed. *Chem. Eng. Comm.*, 194:353-357.
17. Ugwekar, R.P. and G.P. Lakhawat. 2012. Recovery of heavy metal by adsorption using peanut hull. *Int. J. Adv. Eng. Tech.*, 3(3):39-43.
18. Torrado, A. M., *et al.* 2011. Citric acid production from orange peel wastes by solid-state fermentation. *Brazilian J. Microbiol.*, 42(1): 394-409.
19. Rao, M., A.V. Parwate and A.G. Bhole. 2002. Removal of Cr^{6+} and Ni^{2+} from aqueous solution using bagasse and fly ash. *Waste Manage.*, 22(7): +821-830.
20. APHA. 2005. Standard methods for the examination of water and wastewater (21st edn). American Public Health Association, Washington DC.
21. Kowanga, K. D., *et al.* 2016. Kinetic, sorption isotherms, pseudo-first-order model and pseudo-second-order model studies of Cu(II) and Pb(II) using defatted *Moringa oleifera* seed powder. *J. Phytopharmacol.*, 5(2): 71-78.
22. Habib, A., *et al.* 2007. Removal of copper from aqueous solution using orange peel, sawdust and bagasse. *Pakistan J. Anal. Env. Chem.*, 8(1):21-25.
23. Rao, M., A.V. Parwate and P. A. Kadu. 2003. Performance of low-cost adsorbents for the removal of copper and lead using low-cost adsorbent materials. *J. Water Supply: Res. Tech. - AQUA*. 52(1): 49-58.
24. Kumar, P.S., *et al.* 2010. Kinetics and equilibrium studies of Pb^{2+} in removal from aqueous solutions by use of nano-silversol-coated activated carbon. *Brazilian J. Chem. Eng.*, 27(2):339-346.

Integrated Waste Management In Desalination Plants In UAE

Pradeep Kumar V.N., Gabriel Tonga Noweg* and Lau Seng

University Malaysia Sarawak, Institute of Biodiversity and Environmental Conservation, 94300 Kota Samarahan, Sarawak, Malaysia

*Corresponding author, Email : 17010138@siswa.unimas.my ; gtnoweg@unimas.my

Desalination is a process that primarily involves cleanup and purification of sea water for drinking and irrigation purpose. The process is mainly targeted and widely used in the Gulf Cooperation Council country. To concentrate on waste management, it can be stated that it includes all those operations and activities that highly ensure to minimize the waste by adopting effective waste management tools, such as recycling and effectively reusing brine waste. In the current study, it identifies a perfect understanding of integrated waste management in the desalination plant in UAE by analyzing the complete set of the research which includes the various techniques in desalination plant adopted in the UAE and also includes the improvement technique in the desalination process. The study also describes the environmental impact of seawater desalination and also examines the techniques to minimize the negative impact of the desalination plant on the environment. In accession to this, the section also describes the finding and discussion which is grounded on the existing literature which includes a summary of the literature by signifying the major concept. At last, the study also describes the summary of the reassessment that helps to make an effective recommendation for the UAE, so that they will be capable to accomplish the successful implementation of waste management effectively.

KEYWORDS

Desalination, Techniques, Environment, UAE

REFERENCES

1. Dawoud, M. A and M.M. Al Mulla. 2012. Environmental impacts of seawater desalination: Arabian Gulf case study. *Int. J. Env. Sustainability*. 1(3):22-37.
2. Khawaji, A. D., I.K. Kutubkhanah and J.M. Wie. 2008. Advances in seawater desalination technologies. *Desalination*. 221(1-3):47-69.
3. Schenkeveld, M.M., *et al.* 2004. Seawater and brackish water desalination in the Middle East, North Africa and Central Asia: A review of key issues and experiences in six countries. Relatorio tecnico, Banco Mundial, Nimes, France.
4. Sepulveda, A., *et al.* 2010. A review of the environmental fate and effects of hazardous substances released from electrical and electronic equipments during recycling: Examples from China and India. *Env. Impact Assess. Review*. 30(1):28-41.
5. De Nicola, E., *et al.* 2015. Climate change and water scarcity: The case of Saudi Arabia. *Annals Global Health*. 81(3):342-353.
6. Tularam, G.A. and M. Ilahee. 2007. Environmental concerns of desalinating seawater using reverse osmosis. *J. Env. Monitor.*, 9(8):805-813.
7. USEPA. 1976. National pollutant discharge elimination system adjudicatory hearing proceedings: Decisions of the administrator and decisions of the general counsel (vol. 1). Office of General Counsel, US Environmental Protection Agency.
8. Thiel, G. P., *et al.* 2017. Utilization of desalination brine for sodium hydroxide production: Technologies, engineering principles, recovery limits, and future directions. *ACS Sustainable Chem. Eng.*, 5(12):11147–11162.
9. Al-Hogaraty, E.A., Z.S. Rizk and H.K. Garamoon. 2008. Groundwater pollution of the quaternary aquifer in Northern United Arab Emirates. *Water Air Soil Poll.* 190(1-4):323-341.
10. Ahmed, M., *et al.* 2001. Brine disposal from reverse osmosis desalination plants in Oman and the United Arab Emirates. *Desalination*. 133(2):135-147.
11. Vanclay, F. and A.M. Esteves. 2011. New directions in social impact assessment: Conceptual and methodological advances. Edward Elgar Publishing.

12. Stevenson, W.J., M. Hojati and J. Cao. 2007. Operations management (vol. 8). McGraw-Hill/Irwin, Boston.
13. De Vito, C., *et al.* 2011. Reject brines from desalination as possible sources for environmental technologies. In Expanding issues in desalination. InTech. DOI: 10.5772/20302.
14. Sgouridis, S., *et al.* 2016. RE-mapping the UAE's energy transition: An economy-wide assessment of renewable energy options and their policy implications. *Renewable Sustainable Energy Reviews*. 55:1166-1180.
15. Qiblawey, H.M. and F. Banat. 2008. Solar thermal desalination technologies. *Desalination*. 220(1-3):633-644.
16. Perez-Gonzalez, A., *et al.* 2012. State of the art and review on the treatment technologies of water reverse osmosis concentrates. *Water Res.*, 46(2):267-283.
17. Kartam, N., *et al.* 2004. Environmental management of construction and demolition waste in Kuwait. *Waste Manage.*, 24(10):1049-1059.
18. Nair, M., and D. Kumar. 2013. Water desalination and challenges: The Middle East perspective: A review. *Desalination Water Treatment*. 51(10-12):2030-2040.
19. Mezher, T., *et al.* 2011. Techno-economic assessment and environmental impacts of desalination technologies. *Desalination*. 266(1-3):263-273.
20. Mahida, U.N. 1981. Water pollution and disposal of waste water on land. Tata McGraw-Hill, New Delhi.
21. Lattemann, S. and T. Hopner. 2008. Environmental impact and impact assessment of seawater desalination. *Desalination*. 220(1-3):1-15.
22. Islam, M.S. and M. Tanaka. 2004. Impacts of pollution on coastal and marine ecosystems including coastal and marine fisheries and approach for management: A review and synthesis. *Marine Poll. Bulletin*. 48(7-8):624-649.
23. Elshorbagy, W. and A. Alkamali. 2005. Solid waste generation from oil and gas industries in United Arab Emirates. *J. Hazard. Mater.*, 120(1-3):89-99.
24. Giusti, L. 2009. A review of waste management practices and their impact on human health. *Waste Manage.*, 29(8):2227-2239.
25. Shatat, M., M. Worall and S. Riffat. 2013. Opportunities for solar water desalination worldwide. *Sustainable Cities Society*. 9:67-80.
26. Mohamed, A.M.O., M. Maraqa and J. Al Handhaly. 2005. Impact of land disposal of reject brine from desalination plants on soil and groundwater. *Desalination*. 182(1-3):411-433.
27. Khawaji, A. D., I.K. Kutubkhanah and J.M. Wie. 2008. Advances in seawater desalination technologies. *Desalination*. 221(1-3):47-69.

Coir Pith: A Potential Material For Fabrication Of Low-Cost Particle Board

C.R. Sahoo, T.K. Bastia and B.B. Kar*

Kalinga Institute of Industrial Technology (Deemed to be University), Department of Chemistry, School of Applied Sciences, Bhubaneswar - 751 024, Odisha, India

*Corresponding author, Email : bbkarfch@kiit.ac.in; crsevn@gmail.com

The objective of this paper is to utilize waste coir pith, generated from coconut cultivation. During the study, the techno-mechanical properties of the particle board are being tested continuously. In this process, waste coir pith is dried and shredded into a size range within 1 cm. These shredded materials are then ground to smaller particle size in the range of 2-4 mm. The factors under consideration are density (900-1000 kg/m³). The experiments are carried out with variable temperatures in the range of 25-60°C. The boards after fabrication are tested for thickness, moisture content and water absorption. It has been found that all the panel boards are having a much higher density than that of the IS – 3087 standards except urea formaldehyde resin based boards. Similarly, thickness, moisture content, water absorption are within the range of standard IS – 3087. When they are subjected to mechanical parameters, properties such as tensile strength, compressive strength and flexural strength are taken into consideration. The data revealed that as compared to other panels, boards prepared using urea formaldehyde, phenol formaldehyde and melamine formaldehyde resin showed better tensile strength (that is 2 MPa) whereas boards made up of lignin formaldehyde and molasses formaldehyde are of lower tensile strength. Similar results are revealed in terms of compressive and flexural strength as well.

KEYWORDS

Coir pith, Particle board, Urea formaldehyde resin, Lignin formaldehyde

REFERENCES

1. Veigel, S., *et al.* 2012. Particle board and oriented strand board prepared with nanocellulose-reinforced adhesive. *J. Nanomaterials*. DOI: 10.1155/2012/158503.
2. Harshavardhan, A. and L. Muruganandam. 2017. Preparation and characteristic study of particle board from solid waste. *IOP conference series: Mater. Sci. Eng.*, 263(3). DOI: 10.1088/1757-899X/263/3/032005.
3. Gu, K., J. Hung and K. Li. 2013. Biobased adhesives, preparation and evaluation of particle board bonded with a soy flour-based adhesive with a new curing agent. *J. Adhesion Sci. Tech.*, 18-19: 2053-2064.
4. Raleigh, N.C. Review of particle board manufacture and processing. In *Wood technology: Chemical aspects* (chapter 13). Ed M.W. Kelly. pp 220–234.
5. Moubarik, A., *et al.* 2010. Preparation and mechanical characterization of particle board made from maritime pine and glued with bio-adhesives based on corn starch and tannins. *Technologia*. 12(3): 189-197.
6. Naudain, E. A. 1966. Method of preparing a wood particle board. US patent number US3287479A. Available at: <https://patents.google.com/patent/US3287479A/en>.
7. Jin-shu, S., *et al.* 2006. Preparation and properties of waste tea leaves particle board. *Forestry Studies in China*.
8. Farrissey, W. J., *et al.* 1969. Process for preparing particle board. US patent number US4374791A. Available at: <https://patents.google.com/patent/US4374791A/en>.
9. Iwakiri, S., *et al.* 2014. Evaluation of the quality of particle board panels manufactured with wood from *Sequoia sempervirens* and *Pinus taeda*. *CERNE*. 20(2). DOI: 10.1590/01047760.2014.20021524.
10. Kelly, M. W. 1977. Critical literature review of relationships between processing parameters and physical properties of particle board. General technical report. Forest Products Laboratory, Forest Service, U.S. Department of Agriculture.
11. Suleiman, I.Y., *et al.* 2013. Development of eco-friendly particle board composites using rice husk particles and gum arabic. *J. Mater. Sci. Eng. Adv. Tech.*, 7(1):75-91.
12. Muruganandam, L., J. Ranjitha and A. Harshavardhan. 2016. A review report on physical and mechanical properties of particle boards from organic waste. *Int. J. Chem. Tech. Res.*, 9(1): 64-72.
13. Maldas, D. and B.V. Kokta. 1990. Studies on the preparation and properties of particle boards made from bagasse and PVC. I: Effects of operating conditions. *J. Vinyl Additive Tech.*, 12(1): 13-19.

14. Cui, J., *et al.* 2015. Enhancement of mechanical strength of particle board using environmentally friendly pine (*Pinus pinaster* L.) tannin adhesives with cellulose nanofibers. *Annals Forest Sci.*, 72: 27–32.

Removal Of Textile Dye By Titanium Oxide, White Clay And Calcium Oxide

Samir Bekheira, Zohra Dali-Youcef*, Miloud Hamadache and Asma Belaidouni

National Polytechnic School of Oran, Laboratory LABMAT, El Mnaouer Oran, Algeria

*Corresponding author, Email : zohra_dali2002@yahoo.fr; bekheirasamir@gmail.com

The experimental study envisages the elimination of a purple dye KB which is in the discharges of a textile plant ENADITEX of the industrial zone of the Wilaya Oran, Algeria. This dye, like the other dyes, is a pollutant and hardly degradable. Three solid adsorbents: Titanium oxide, white clay and calcium oxide were used to remove the organic dye KB (cationic type) used in the textile industry. The kinetic study of the purple dye adsorption process showed that the removal using Titanium oxide, white clay and calcium oxide attained was 87%, 80% and 86%, respectively for respective contact times of 200 min, 60 min, 50 min and that the adsorption of the purple dye by the three solid materials was well described by the pseudo-second order model while the thermodynamic study revealed the adsorption process was physical. On the other hand, the study of adsorption isotherms using different classical models showed that adsorption can be described by the Freundlich and Temkin isotherm models.

KEYWORDS

Titanium oxide, White clay, Calcium oxide, Purple dye KB, Adsorption isotherms

REFERENCES

- Ahmad, A., *et al.* 2009. Scavenging behaviour of meranti sawdust in the removal of Methylene Blue from aqueous solution. *J. Hazard. Mater.*, 170:357-365.
- Khelifi, E., *et al.* 2008. Aerobic decolourization of the indigo dye-containing textile wastewater using continuous combined bioreactors. *J. Hazard. Mater.*, 152:683-689.
- Daneshvar, N., *et al.* 2003. Photocatalytic degradation of azo dye Acid Red 14 in water: Investigation of the effect of operational parameters. *J. Photochem. Photobiol A: Chem.*, 157:111-116.
- Sami, G. and B. Mohamed. 2013. Kinetic study of the adsorption of Congo Red on a bentonite. *J. Water Sci.*, 26:39-50.
- Mohamed, L., *et al.* 2015. Kinetic and thermodynamic study of the adsorption of monoazo dyes on polyaniline. *J. Mater. Env.*, 6:1049-1059.
- Paya, J., *et al.* 2000. Mechanical treatment of fly ashes: Part IV. Strength development of ground fly ash-cement mortars cured at different temperatures. *Cement Concrete*. 30:543-551.
- Belkacem, B. and Y.N. Aicha. 2009. Elimination of acid dyes in aqueous solution by bentonite and kaolin. *Comptes Rendus Chimie*. 12:762-771.
- Fei-fei, L., *et al.* 2013. Adsorption of natural organic matter analogues by multi-walled carbon nanotubes: Comparison with powdered activated carbon. *Chem. Eng. J.*, 219:450-458.
- Gomez, V., *et al.* 2007. Kinetic and adsorption study of acid dye removal using activated carbon. *Chemosphere*. 69:1151-1158.
- Kumar, M. and R. Tamilarasan. 2014. Removal of Victoria Blue using *Prosopis juliflora* bark carbon: Kinetics and thermodynamic modeling studies. *J. Mater. Env.*, 5(2):510-519.
- Elbariji, S., *et al.* 2006. Treatment and upgrading of wood by-products. Application to the removal of industrial dyes. *Comptes Rendus Chimie*. 9: 1314-1321.
- Monarrez, I.M. 2004. Pesticide retention in soil buffer devices, herbs and wood and role of organic matter. Ph.D. Thesis. Grignon University, Paris, France.
- Iqkhllass, M.T. 2012. Study of chromium VI removal by activated alumina by cross ionic dialysis. Ph.D. Thesis. Est University, Paris, France.
- Lenoble, V. 2003. Arsenic removal for drinking water production: Chemical oxidation and adsorption on innovative solid substrates. Ph.D. Thesis. Limoges University, France.
- Chafai, H., *et al.* 2013. Adsorption of sodium salicylate and Cr (VI) by polypyrrole. *J. Mater. Env. Sci.*, 4(2): 285-292.16.

16. Mohamed, L., *et al.* 2015. Kinetic and thermodynamic study of the adsorption of monoazo dyes on polyaniline. *J. Mater. Env. Sci.*, 6(4): 1049-1059.
17. Belmouden, M., *et al.* 2001. Removal of 2-4 dichloro phenoxyacetic acid from aqueous solution by adsorption on activated carbon: A kinetic study. *Annales de Chimie Sci. des Matériaux.*, 26(2):79-85.
18. Oguz, E. 2005. Adsorption characteristics and the kinetics of the Cr (VI) on the *Thuja orientalis*. *J. Colloid Surfaces A: Physicochem. Eng. Aspects.* 252:121-128.
19. Romero-Gonzalez, J., *et al.* 2005. Determination of thermodynamic parameters of Cr (VI) adsorption from aqueous solution onto *Agave lechuguilla* biomass. *J. Chem. Thermodynamics.* 37:343-347.

Heavy Metal Uptake By Wheat And Mustard Crops Grown in Surrounding Of A Coal-Fired Thermal Power Plant, Bathinda (Punjab)

Savita Verma and Anju*

Chaudhary Devi Lal University, Department of Energy and Environmental Sciences, Sirsa - 125 055, Haryana, India

*Corresponding author, Email : anjumalik@cdlu.ac.in; drsavi13@gmail.com

Plants show natural ability to absorb and accumulate trace metals from soil. The aim of the present research paper is to study the accumulation and distribution pattern of some heavy metals, namely Cr, Cu, Pb, Ni and Zn in different parts (root, stem, leaf and seed) of wheat and mustard crops, grown in fields around Guru Nanak Dev Thermal Power Plant Bathinda, Punjab. For this purpose, the bioconcentration factor (BCF) and translocation factor (TF) of metals between the different plant parts of both the crops and the soils were studied. Findings of the present study indicate that all the studied metals have primarily accumulated in roots of both the crops. In both crops, the concentrations of all the studied heavy metals in plant tissues were diminishing from the root to the aerial parts. The levels of all the trace elements were higher in the mustard crop than the wheat crop. The concentrations of all the metals in both crops were within safe limits of Indian standards for plants.

KEYWORDS

Wheat, Mustard, Heavy metal, Bioconcentration factor, Translocation factor

REFERENCES

1. Logan, T. J. and R. L. Chaney. 1983. Utilization of municipal wastewater and sludge on land - Metals. In Proceedings of the Workshop on Utilization of municipal wastewater and sludge on land. Ed A. L. Page, *et al.*, University of California, Riverside, California. pp 235-323.
2. Abulude, F.O. and H. Adesoje. 2006. Characterization of heavy metal pollution around Cassava processing factory using atomic absorption spectrophotometer. *Res. J. Appl. Sci.*, 1(1-4):16-18.
3. Cheng, S.P. 2003. Heavy metal pollution in China: Origin, pattern and control. *Env. Sci. Poll. Res.*, 10(3):192-198.
4. Kabata-Pendias, A. and A.B. Mukherjee. 2007. Trace elements from soil to human. Springer-Verlag Berlin Heidelberg.
5. Gupta, U. and S. Gupta. 1998. Trace element toxicity relationships to crop production and livestock and human health: Implications for management. *Communications Soil Sci. Plant Analysis*. 29:1491-1522.
6. McBride, M.B. 2007. Trace metals and sulphur in soils and forage of a chronic wasting disease locus. *Env. Chem.*, 4:134-139.
7. Monika, D.P. and S.K. Katarzyna. 2004. Histopathological changes in the liver, kidneys and testes of bank voles environmentally exposed to heavy metal emissions from the steelworks and zinc smelter in Poland. *Env. Res.*, 96:72-78.
8. Ernst, W. 1982. Heavy metal plants. In H. Kinzel. Ed Plants ecology and mineral metabolism. Ulmer, Stuttgart. pp 472-506.
9. Hall, J. L. and L.E. Williams. 2003. Transition metal transporters in plants. *J. Experimental Botany*. 54(393): 2601-2613. DOI: 10.1093/jxb/erg303.
10. Tangahu, B.V., *et al.* 2011. A review on heavy metals (As, Pb and Hg) uptake by plants through phytoremediation. *Int. J. Chem. Eng.*, 1-31. DOI: 10.1155/2011/939161.
11. Sengupta, S., *et al.* 2011. Heavy metal contamination in leaves of *Mangifera indica* around a coal fired thermal power plant in India. *J. Ecol. Nat. Env.*, 3(14):446-454.
12. Cordes, K. B., *et al.* 2000. Uptake of Cd, Cu, Ni and Zn by the water hyacinth, *Eichhornia crassipes* (Mart.) solms from pulverised fuel ash (PFA) leachates and slurries. *Env. Geochem. Health*. 22 (4):297-316.

13. Singh, R., *et al.* 2010. Accumulation and translocation of heavy metals in soil and plants from fly ash contaminated area. *J. Env. Bio.*, 31:421-430.
14. Sjostrom, A.E., *et al.* 2008. Degradation and plant uptake of nonylphenol and nonylphenol-12-ethoxylate in four contrasting agricultural soils. *Env. Poll.*, 3:1284-1289.
15. Sipter, E., *et al.* 2008. Site specific risk assessment in contaminated vegetable gardens. *Chemosphere*. 71:1301-1307.
16. Rico, M.I., *et al.* 2009. Manganese and zinc in acidic agricultural soils from Central Spain: Distribution and phytoavailability prediction with chemical extraction tests. *Soil Sci.*, 174:94-104. DOI: 10.1097/SS.0b013e3181975058.
17. Zhang, X., *et al.* 2010. Tolerance and accumulation characteristics of cadmium in *Amaranthus hybridus* L. *J. Hazard. Mater.*, 180(1-3):303-308.
18. Ma, L.Q., *et al.* 2001. A fern that hyperaccumulates arsenic. *Nature*. 409(6820):579.
19. Sauerbeck, D. R. 1991. Plant, element and soil properties governing uptake and availability of heavy metals derived from sewage-sludge. *Water Air Soil Poll.*, 57-58:227-237.
20. Cui, Y.J., *et al.* 2004. Transfer of metals from soil to vegetables in an area near a smelter in Nanning, China. *Env. Int.*, 30:785-791.
21. Chojnacka, K., *et al.* 2005. Bioavailability of heavy metals from polluted soils to plants. *Sci. Total Env.*, 337:175-182.
22. Intawongse, M. and J. R. Dean. 2006. Uptake of heavy metals by vegetable plants grown on contaminated soil and their bioavailability in the human gastro-intestinal tract. *Food Additives Contaminants*. 23:36-48.
23. Ramadas, S., T.M. K. Kumar and G.P. Singh. 2019. Wheat production in India: Trends and prospects. In Recent advances in grain crops research. Ed Farooq Shah, *et al. IntechOpen*. DOI: 10.5772/intechopen.86341.
24. Verma, H. and J. Dawson. 2018. Yield and economics of mustard (*Brassica campestris* L.) as influenced by sowing methods and levels of sulphur and boron. *Int. J. Current Microbiol. Appl. Sci.*, 7(1):380-386. DOI: 10.20546/ijcmas.2018.701.043.
25. Subhashini, V. and V.V.S. Swamy. 2013. Phytoremediation of zinc contaminated soils by *Physalis minima* Linn. *American Int. J. Res. Formal, Appl. Nat. Sci.*, 2:4488-4492.
26. Kozakova, E., M. Vackova and A. Uvackova. 1991. Utilization of voltamperometric methods and AAS methods for the determination of certified values of selected heavy-metals in reference materials of fly ashes. *Chemickelisty*. 85(1):83-93.
27. APHA. 2005. Standard methods for the examination of water and wastewater (21st edn). American Public Health Association, New York, USA.
28. Meyers, D.E., *et al.* 2008. Uptake and localisation of lead in the root system of *Brassica juncea*. *Env. Poll.*, 153(2):323-332. DOI: 10.1016/j.envpol.2007.08.029.
29. Zaier, H., *et al.* 2010. Effects of EDTA on phytoextraction of heavy metals (Zn, Mn and Pb) from sludge-amended soil with *Brassica napus*. *Bioresour. Tech.*, 101(11):3978-3983.
30. Ali, A., *et al.* 2017. Role of *Streptomyces pactum* in phytoremediation of trace elements by *Brassica juncea* in mine polluted soils. *Ecotoxicol. Env. Safety*. 144:387-395. DOI: 10.1016/j.ecoenv.2017.06.046.
31. Shanker, A.K., *et al.* 2005. Chromium toxicity in plants. *Env. Int.*, 31:379-753.
32. Asdeo, A. 2014. Toxic metal contamination of staple crops (wheat and millet) in Periurban area of Western Rajasthan. *Int. Ref. J. Eng. Sci.*, 3(4):8-18.
33. FAO/WHO. 1984. Contaminants. In Codex Alimentarius (vol. xvii, edn 1). FAO/WHO, Codex Alimentarius Commission, Rome.
34. Awashthi, S.K. 2000. Prevention of food adulteration act no. 37 of 1954. Central and state rules as amended for 1999, Ashoka Law House, New Delhi.
35. Avudainayagam, S., *et al.* 2003. Chemistry of chromium in soils with emphasis on tannery waste sites. *Reviews Env. Contamination Toxicol.*, 178:53-91.
36. Lopez-Luna, J., *et al.* 2009. Toxicity assessment of soil amended with tannery sludge, trivalent chromium and hexavalent chromium, using wheat, oat and sorghum plants. *J. Hazard. Mater.*, 163:829-834.
37. Gupta, A.K. and S. Sinha. 2007. Assessment of single extraction methods for the prediction of bioavailability of metals to *Brassica juncea* L. Czern. (var. Vaibhav) grown on tannery waste contaminated soil. *J. Hazard. Mater.*, 149(1):144-150.
38. Lopez-Luna, M.C., *et al.* 2012. Fractionation and availability of heavy metals in tannery sludge-amended soil and toxicity assessment on the fully-grown *Phaseolus vulgaris* cultivars. *J. Env. Sci. Health*. 47(3):405-419.
39. Barman, S.C., *et al.* 2000. Distribution of heavy metals in wheat, mustard, and weed grown in field irrigated with industrial effluents. *Bull. Env. Contam. Toxicol.*, 64:489-496. Doi: 10.1007/s0012800 00030.

40. Kim, I.S., *et al.* 2003. Investigation of heavy metal accumulation in *Polygonum thunbergii* for phytoextraction. *Env. Poll.*, 126:235–243.
41. Alloway, B.J. 1995. Heavy metals in soils (2nd edn). Springer Netherlands. pp 38–57.
42. Jarvis, S.C., L.H.P. Jones and M.J. Hopper. 1976. Cadmium uptake from solution by plants and its transport from roots to shoots. *Plant Soil*. 44:179-191.
43. Huang, M., *et al.* 2008. Heavy metals in wheat grain: Assessment of potential health risk for inhabitants in Kunshan, China. *Sci. Total Env.*, 405:54–61.
44. Kutrowska, A., *et al.* 2017. Effects of binary metal combinations on zinc, copper, cadmium and lead uptake and distribution in *Brassica juncea*. *J. Trace Element Medicine Biol.*, 44:32–39.
45. Wang, X. S. and Y. Qin. 2005. Accumulation and sources of heavy metals in urban top soils: A case study from the city of Xuzhou, China. *Env. Geol.*, 48:101-107.
46. Pueyo, M., *et al.* 2003. Prediction of trace element mobility in contaminated soils by sequential extraction. *J. Env. Quality*. 32:2054-2066.
47. Lorestani, B., M. Cheraghi and N. Yousefi. 2011. Accumulation of Pb, Fe, Mn, Cu and Zn in plants and choice of hyperaccumulator plant in the industrial town of Vian, Iran. *Archives Biol. Sci.*, 63:739–745.
48. Singh, S. and S. Sinha. 2005. Accumulation of metals and its effects in *Brassica juncea* (L.) Czern. (cv. Rohini) grown on various amendments of tannery waste. *J. Ecotoxicol. Env. Safety*. 62(1):118-127.
49. Kisku, G.C., *et al.* 2011. Uptake and accumulation of potentially toxic metals (Zn, Cu and Pb) in soils and plants of Durgapur industrial belt. *J. Env. Biol.*, 32:831–838.

Socio-Economic Scenario Of Scrapers And Their Role In Municipal Solid Waste Management In City Of Varanasi

Vijai Krishna*, Dharmendra Kumar, Rajni Srivastava, Himanshu Kumar and Anil Kumar Pandey

Banaras Hindu University, Institute of Environment and Sustainable Development, Varanasi - 221 005, U.P., India

*Corresponding author, Email : vijaikrishna@bhu.ac.in; vdharmendra493@gmail.com

Scrapers or scrap dealers or scrap traders play an important role in the municipal solid waste management of a city, like collection, storage, sorting, reuse and recycling of waste. Scrapping has become a livelihood practice of a large group of people especially in developing countries, where government authorities are not much efficient in waste management. Scrapers not only clean the cities but also help the city authorities by doing so. But their socio-economic condition is not well. In this research paper the results of a survey have been discussed to assess the socio-economic scenario of scrapers in the city of Varanasi of Uttar Pradesh state in India, like education, health, entertainment, help from NGOs, police harassment and their practice of waste collection, storage, sorting, selling, etc. The results of the survey clearly indicate that the socio-economic condition of these scrapers is very bad and need the attention of government and waste management authorities for the improvement of them.

KEYWORDS

Scrapers, Municipal solid waste management, Socio-economic scenario, Varanasi

REFERENCES

1. Srivastava, R., V. Krishna and I. Sonkar. 2014. Characterization and management of municipal solid waste: A case study of Varanasi city. *Int. J. Current Res. Academic Review*. 2(8): 10-16.
2. Kaushal, R.K., G.K. Varghese and M. Chabukdhara. 2012. Municipal solid waste management in India - Current state and future challenges: A review. *Int. J. Eng. Sci. Tech.*, 4(4): 1473-1489.
3. Sharholly, M., et al. 2007. Municipal solid waste characteristics and management in Allahabad. *J. Waste Manage.*, 27: 490-496.
4. Krishna, V. and S. Chaurasia. 2017. Assessment of potential of energy recovery from municipal solid waste of Allahabad city. *Int. J. Appl. Res. Tech.*, 2(3): 165-171.
5. Krishna, V. and S. Chaurasia. 2016. Assessment of socio-economic condition and role of rag pickers in municipal solid waste management in Allahabad city (U.P.). *Int. J. Appl. Res. Tech.*, 1(1): 13-20.
6. Krishna, V., H. Kumar and S. Chaurasia. 2019. Socio-economic scenario of rag pickers and their role in municipal solid waste management of city of Varanasi. *J. Emerging Tech. Innovative Res.*, 6(1): 626-641.
7. Idris, A., B. Inane and M.N. Hassan. 2004. Overview of waste disposal and landfills dumps in Asian countries. *Mater. Cycles Waste Manage.*, 16:104-110.
8. Kumar, S. and S.A. Gaikwad. 2004. Municipal solid waste management in Indian urban centres. In An approach for betterment, urban development debates in the new millennium. Ed K. R. Gupta. Atlantic Publications and Distributors, New Delhi. pp 100-111.
9. Saxena, S., R.K. Srivastava and A.B. Samaddar. 2010. Sustainable waste management issues in India. *J. Soil Water Sci.*, 3(1): 72-90.
10. Zhu, D., et al. 2008. Improving solid waste management in India - A sourcebook for policymakers and practitioners. World Bank Institute, WBI Development Studies, The World Bank.
11. Gidde, M. R., V.V. Todkar and K.K. Kokate. 2008. Municipal solid waste management in emerging mega cities: A case study of Pune city. Indo Italian Conference on Green and clean environment proceeding, Pune.
12. Rathi, S. 2007. Optimization model for integrated municipal solid waste management in Mumbai. *Env. Develop. Economics*. 12(1): 105-121.
13. Syed, S. 2006. Solid and liquid waste management. *Emirates J. Eng. Res.*, 11(2): 19-36.
14. Rajkumar, N., T. Subramani and L. Elango. 2010. Groundwater contamination due to municipal solid waste disposal - A GIS based study in Erode city. *Int. J. Env. Sci.*, 1(1): 39-55.

15. Rachel, O. A., *et al.* 2009. Municipal solid waste management in developed and developing countries - Japan and Nigeria as case studies. Available at: [http://www.geo.civil.ibaraki.ac.jp/komine/mypapers/JGSPaper/2009/JGS2009\(973\) Rachel.pdf](http://www.geo.civil.ibaraki.ac.jp/komine/mypapers/JGSPaper/2009/JGS2009(973) Rachel.pdf).
16. Ogu, V.I. 2000. Private sector participation and municipal waste management in Benin city, Nigeria. *Env. Urban.* 12(2): 103-117.
17. Jha, A.K., *et al.* 2011. Sustainable municipal solid waste management in low income group of cities: A review. *Tropical Ecol.*, 52(1): 123-131.
18. Heart, S. 2009. Electronic waste: An emerging issue in solid waste management in Australia. *Int. J. Env. Waste Manage.*, 3(1/2): 120-134.
19. Tchobanoglous, G. H. T. and S.A. Vigil. 1993. Integrated solid waste management: Engineering principles and management issues. McGraw-Hill, New York.
20. Krishna, V. and S. Chaurasia. 2017. Aspects of municipal solid waste management in Allahabad city: A questioner survey of the citizens. *IOSR J. Env. Sci. Toxicol. Food Tech.*, 11(2) (1): 11-16.
21. Ojha, V.K., V. Tripathi and V. Krishna. 2014. Role of rag pickers in solid waste management: A case study of Mirzapur, U.P. *Int. J. Env. Tech. Manage.*, 17(2/3/4): 134-142.
22. Rochman, C. M., *et al.* 2015. Anthropogenic debris in seafood: Plastic debris and fibers from textiles in fish and bivalves sold for human consumption. *Sci. Reports.* 5. DOI: 10.1038/srep14340.
23. Marthanadan, K.V. 2007. Urbanization and municipal solid waste management. *J. Eng. Appl. Sci.*, 2(6): 977-988.
24. Chattopadhyay, S., A. Dutta and S. Ray. 2007. Sustainable municipal solid waste management for the city of Kolkata. International Conference on Civil engineering in the new millennium: Opportunities and challenges. Bengal Engineering and Science University, Shibpur.
25. Varanasi: About the city. Available at: Varanasi.nic.in.

Experimental Evaluation Of *Solanum tuberosum* Peel As Bioadsorbent For Removal Of Nickel And Chromium Ions From Aqueous Solution

Harendra K. Sharma^{1*}, Irfan Rashid Sofi¹ and Laxmi Dubey²

1. Jiwaji University, School of Studies in Environmental Science, Gwalior - 474 001, M.P., India

2. SMS Govt. Model Science College, Department of Botany, Gwalior - 474 009, M.P., India

*Corresponding author, Email : drhksharmagwl@gmail.com

The ability of potato peel to act as biosorption medium in the removal of metal ions from aqueous solution was investigated. The adsorbent removal efficiency was determined as a function of contact time, initial metal ion concentration, pH and temperature. Potato peel was characterized by scanning electron microscopy (SEM) and Fourier transformed infrared spectroscopy (FTIR). FTIR results showed that the functional groups providing the active sites are carboxyl, hydroxyl and carbonyl groups. Results from the biosorption experiment showed that potato peel displays a good binding affinity with metals ions [Cr(VI) and Ni(II)] in the aqueous solution. The equilibrium profile fits with the Langmuir isotherm with a coefficient values of 0.93 and 0.99 for Ni(II) and Cr(VI) and Freundlich isotherm having coefficient values of 0.97 and 0.99 for Ni(II) and Cr(VI), respectively. The potato peel was found to be highly efficient in the removal of selected heavy metal ions.

KEYWORDS

Biosorption, Potato peel, Heavy metals, Aqueous solution, Adsorption isotherm

REFERENCES

1. Haware, D.J. and H.P. Pramod. 2011. Determination of specific heavy metals in fruit juices using atomic absorption spectrophotometer. *Int. J. Res. Chem. Env.*, 4: 163-168.
2. Zhang, X., et al. 2011. Assessing the toxicity of naphthenic acids using a microbial genome wide live cell reporter array system. *Env. Sci. Tech.*, 45(5):1984-1991.
3. Prakasham, R.S., et al. 1999. Biosorption of chromium VI by free and immobilized *Rhizopus arrhizus*. *Env. Poll.*, 104: 421-427.
4. WHO. 2014. World cancer report. Ed B. W. Stewart and C. P. Wild. International Agency for Research on Cancer, World Health Organization, Geneva.
5. Dubey, L., H.K. Sharma and R.K. Khare. 2017. Removal of zinc and lead from aqueous solution using low cost bioadsorbent *Pennisetum glaucum* (Bajara) husk. *Adv. Biores.*, 8:188-196.
6. Bai, R.S. and T.E. Abraham. 2003. Studies on chromium (VI) adsorption-desorption using immobilized fungal biomass. *Bioresour. Tech.*, 87: 17-26.
7. Bailey, S.E., et al. 1999. A review of potentially low-cost sorbents for heavy metals. *Water Resour. Res.*, 33: 2469-2479.
8. Uzun, H., et al. 2002. Biosorption of chromium (VI) from aqueous solution by cone biomass of *Pinus sylvestris*. *Bioresour. Tech.*, 85:155-158.
9. Zuo, X.J., et al. 2012. Biosorption of copper, zinc and cadmium using sodium hydroxide immersed *Cymbopogon schoenanthus* L, Spreng (lemon grass). *Ecol. Eng.*, 49: 186-18.
10. Bingol, A., et al. 2004. Removal of chromate anions from aqueous stream by a cationic surfactant modified yeast. *Bioresour. Tech.*, 94: 245-249.
11. Sciban, M.B., M.T. Klasnja and M.G. Antov. 2011. Study of the biosorption of different heavy metal ions onto Kraft lignin. *Ecol. Eng.*, 37: 2092-2095.
12. Singh, L., et al. 2012. Effective removal of Cu²⁺ ions from aqueous medium using alginate as biosorbent. *Ecol. Eng.*, 38: 119-124.

13. Areco, M.M., *et al.* 2012. Biosorption of Cu(II), Zn(II), Cd(II) and Pb(II) by dead biomasses of green alga *Ulva lactuca* and the development of a sustainable matrix for adsorption implementation. *J. Hazard. Mater.*, 213-214:123-132.
14. Bulgariu, D. and L. Bulgariu. 2012. Equilibrium and kinetics studies of heavy metal ions biosorption on green algae waste biomass. *Bioresour. Tech.*, 103: 489-493.
15. Cabatingan, L.K., *et al.* 2001. Potential of biosorption for the recovery of chromate in industrial wastewaters. *Ind. Eng. Chem. Res.*, 40: 2302-2309.
16. Lee, Y.C. and S.P. Chang. 2011. The biosorption of heavy metals from aqueous solution by *Spirogyra* and *Cladophora* filamentous macroalgae. *Bioresour. Tech.*, 102: 5297-5304.
17. Iqbal, M., A. Saeed and S.I. Zafar. 2008. FTIR spectrophotometry, kinetics and adsorption isotherm modeling, ion exchange and EDX analysis for understanding the mechanism of Cd²⁺ and Pb²⁺ removal by mango peel waste. *J. Hazard. Mater.*, 164: 161-171.
18. Sharma, H.K., I.R. Sofi and K.A. Wani. 2018. Low cost absorbents, techniques and heavy metal removal efficiency. In *Biostimulation remediation technologies for groundwater contaminants*. IGI Global. pp 50-79.
19. Helgi Library. 2019. Potato consumption per capita in India. Helgi Analytics, Czech Republic.
20. Pratt, D.Y., L.D. Wilson and J.A. Kozinski. 2013. Preparation and sorption studies of glutaraldehyde cross-linked chitosan copolymers. *J. Colloid Interface Sci.*, 395:205-211.
21. Pandharipande, S.L. and R.P. Kalnake. 2013. Tama rind fruit shell adsorbent synthesis, characterization and adsorption studies for removal of Cr(VI) ions from aqueous solution. *Int. J. Emerging Res. Eng. Sci.*, 4: 83-89.
22. Mutongo, F., O. Kuipa and P.G. Kuipa. 2014. Removal of Cr(VI) from aqueous solutions using powder of potato peelings as a low cost sorbent. *Bioinorganic Chem. Applications*. DOI: 10.1155/2014/973153.
23. El-Said, A.G. 2010. Biosorption of Pb(II) ions from aqueous solutions onto rice husk and its ash. *American J. Sci.*, 6: 143-150.

Treatment Of Slaughterhouse Wastewater By Upflow Anaerobic Hybrid Reactor At Pilot Scale

Rohit Patel^{1*}, Hemant Kumar², Shobha Ram² and Ashish Kumar Sisodia²

1. Indian Institute of Technology, Indian School of Mines, Dhanbad - 826 004, Jharkhand, India

2. Gautam Buddha University, Greater Noida - 201 308, Uttar Pradesh, India

*Corresponding author, Email : rohitpatel7@gmail.com

The wastewater discharged by slaughterhouse industries is categorized mainly by high COD, high BOD, high suspended and complex mixture of fats, proteins and fibres which needs systematic treatment before the disposal. In the present study, the performance of 3.26 L upflows anaerobic hybrid reactor treating the slaughterhouse wastewater with polyvinyl chloride (PVC) uses as filter media was investigated. For seeding, the reactor is inoculated with activated sludge from a upflow anaerobic sludge blanket (UASB) reactor treating the slaughterhouse wastewater. After seeding, the reactor with maintained, inlet flow of 135.72 mL/hr is loaded with different dilution ratios of the sample (1/10, 1/6, 1/4 and 1/1 or undiluted) with different COD loadings (9.281 kg.COD/m³ day, 10.149 kg.COD/m³ day, 9.263 kg.COD/m³ day and 11.210 kg.COD/m³ day) with HRT of 24 hr and achieved maximum COD removal efficiency upto 73.32% at COD loading of 11.210 kg.COD/m³ day. Alongwith the COD, other required parameters, such as phosphorus, BOD, TSS, VSS and volatile acid as acetic acid are also found out to check the efficiency of the reactor.

KEYWORDS

Slaughterhouse waste, Anaerobic digestion, Anaerobic filter, UASB reactor, Chemical oxygen demand

REFERENCES

- Banu, J.R., S. Kaliappan and I.T. Yeom. 2007. Treatment of domestic wastewater using upflow anaerobic sludge blanket reactor. *Int. J. Env. Sci. Tech.*, 4(3): 363-370.
- Masse, D.I. and L. Masse. 2000. Treatment of slaughterhouse wastewater in anaerobic sequencing batch reactors. *Canadian Agric. Eng.*, 42(3): 131-137.
- Jhung, J. K. and E. Choi. 1995. Relative study of USB and anaerobic fixed film reactor with the development of sludge granulation. *Water Res.*, 29(1): 271-277.
- Juang and D.F. Chiou. 2007. Population structures in activated sludge earlier and after the application of synthetic polymer. *Int. J. Env. Sci. Tech.*, 4(1): 119-125.
- Liu, R. R., *et al.* 2010. Hybrid anaerobic baffled reactor for the treatment of resizing wastewater. *Int. J. Env. Sci. Tech.*, 7(1): 111-118.
- Jawed, M. and V. Tare. 1999. Microbial configuration assessment of anaerobic biomass through methanogenic activity tests. *Water SA.*, 25(3): 345-350.
- Jawed, M. and V. Tare. 2000. Post-mortem assessment and study of anaerobic filters. *Biores. Tech.*, 72(1): 75-84.
- Ruiz, I., *et al.* 1997. Treatment of slaughterhouse wastewater in an anaerobic filter and UASB reactors. *Biores. Tech.*, 60(3): 251-258.
- Delpozo, R., V. Diez and S. Beltran. 2000. Anaerobic pre-treatment of slaughterhouse wastewater with the help of fixed-film reactors. *Biores. Tech.*, 71(2): 143-149.
- Chan, Y.J., *et al.* 2009. Anaerobic-aerobic treatment of industrial and municipal wastewater: A review. *Chem. Eng. J.*, 155: 1-18.
- Debik, E. and T. Coskun. 2009. Use of the static granular bed reactor (SGBR) with anaerobic sludge to treat poultry slaughterhouse wastewater and kinetic modeling. *Bioresour. Tech.*, 100: 2777-2782.
- Tawfik, A. and H. El-Kamah. 2012. Treatment of fruit-juice industry wastewater in a two-stage anaerobic hybrid (AH) reactor system followed by a sequencing batch reactor (SBR). *Env. Tech.*, 33: 429-436.
- Braguglia, C.M., A. Gianico and G. Mininni. 2012. Comparison between ozone and ultrasound disintegration on sludge anaerobic digestion. *J. Env. Manage.*, 95: S139-S143.

14. Callegari, A., V. Torretta and A.G. Capodaglio. 2013. Introductory test application of biological desulfonation in anaerobic digestors from pig farms. *Env. Eng. Manage. J.*, 12(4): 815-819.
15. Rada, E.C., M. Ragazzi and V. Torretta. 2013. Laboratory-scale anaerobic sequencing batch reactor for treatment of stillage from fruit distillation. *Water Sci. Tech.*, 67: 1068-1074.
16. Rada, E.C. and M. Ragazzi. 2008. Critical analysis of PCDD/F emissions from anaerobic digestion. *Water Sci. Tech.*, 58: 1721-1725.
17. Torretta, C. F., M. Leonardi and G. Ruggeri. 2012. Energy regaining from sludge and sustainable development: A Tanzanian case study. *Sustainability*. 4: 2661-2672.
18. Behera, S.K., E. R. Rene and D.V.S. Murthy. 2007. Performance of up-flow anoxic bioreactor for wastewater treatment. *Int. J. Env. Sci. Tech.*, 4(2): 247-252.
19. Delpozo, R. and V. Diez. 2005. Integrated anaerobic-aerobic fixed-film reactor for slaughterhouse wastewater treatment. *Water Res.*, 39: 1114–1122.
20. Hulshoff-Pol, L.W., *et al.* 1983. Granulation in UASB reactors. *Water Sci. Tech.*, 15(8/9): 291-304.
21. Grotenhuis, J.T.C., *et al.* 1991. Role of substrate concentration in particle size circulation of methanogenic granular sludge in UASB reactors. *Water Res.*, 25(1): 21-27.
22. Borja, R. and C.J. Banks. 1995. Assessment of an anaerobic filter and anaerobic fluidized bed reactor discussing palm oil mill effluent. *Process Biochem.* 30(6): 511-521.
23. Bicheldey, T.K. and E. Latushkina. 2010. Biogas release prognosis at the landfills. *Int. J. Env. Sci. Tech.*, 7(4): 623-628.
24. Iscen, F.C., S. Ilhan and M.E. Yildirim. 2007. Treatment of cake fabrication wastewater in upflow anaerobic packed bed reactors. *Int. J. Natural Eng. Sci.*, 1(3): 75-80.
25. Guerrero, L., *et al.* 1997. Treatment of saline wastewaters from fish meal factories in an anaerobic filter under great ammonia concentrations. *Biores. Tech.*, 61(1): 69-78.

Improvement In Quality Of AMD Contaminated Water By Treatment With Limestone And Filtration Through Activated Charcoal And Sand Filter

Ribha I. Passah and O. P. Singh*

North-Eastern Hill University, Department of Environmental Studies, Shillong - 793 022

*Corresponding author, Email : opsinghnehu@gmail.com; passahribha@gmail.com

Unscientific coal mining in Jaintia Hills, Meghalaya has severely degraded the environment of the area. The water of the area has become highly acidic and contaminated with different impurities making the water unsuitable for domestic and irrigation uses leading to scarcity of clean water in the mining area. Acid mine drainage (AMD) has been recognized as the main cause of water pollution. This calls for an urgent action to improve the quality of water to ease the water stress in the area. Laboratory experiments involving treatment of acidic water using locally available limestone and subsequent filtration through the layers of activated charcoal and sand yielded promising results towards the improvement of water quality. Treatment with limestone raised the pH from 2.8-7.31, well within the acceptable limit. Subsequent filtration improved the water quality further by bringing the values of calcium, turbidity, total dissolved solids (TDS), alkalinity and metals, such as zinc and aluminium within the acceptable limits prescribed by Bureau of Indian Standards. The promising results of the present study can be used for designing suitable filtration units for domestic uses as well as for treatment of stream water for irrigation and ecological purposes.

KEYWORDS

Acid mine drainage, Coal mining, Improvement in water quality, Neutralization by limestone, Filtration with charcoal and sand

REFERENCES

1. Directorate of Mineral Resources. 1985. Technical report of the Directorate of Mineral Resour., Government of Meghalaya, Shillong, Meghalaya.
2. Swer, S. and O. P. Singh. 2003. Coal mining impacting water quality and aquatic biodiversity in Jaintia hills district of Meghalaya. *Envis Bulletin-Himalayan Ecol.*, 11:26-33.
3. Swer, S. and O. P. Singh. 2004a. Status of water quality in coal mining areas of Meghalaya. National Seminar on Env. eng. with special emphasis on mining env. Proceedings, pp 173-181.
4. Blahwar, B., et al. 2012. Use of high-resolution satellite imagery for investigating acid mine drainage from artisanal coal mining in north-eastern India. *Geocarto. Int.*, 27(3):231-247.
5. Chabukdhara, M. and O. P. Singh. 2016. Coal mining in northeast India: An overview of env. issues and treatment approaches. *Int. J. Coal Sci. and Tech.*, 3(2):87-96.
6. Swer, S. and O. P. Singh. 2004b. Water quality, availability and aquatic life affected by coal mining in ecologically sensitive areas of Meghalaya. 3rd National Seminar on Inland water resour. and env., Thiruvananthapuram, Kerala. Proceedings, pp 102-108.
7. Somendro, T. and O. P. Singh. 2015. Analysis of landuse/land covers dynamics using remote sensing and GIS techniques: A case study of Jaintia Hills, Meghalaya. *Int. J. Current Res.*, 7:11873-11879.
8. Lamare R. E., et al. 2019. Landuse/ land cover change in East Jaintia Hills, Meghalaya in relation to limestone mining. *Env. and Ecol.*, 37(3A):886-893.
9. Wood, C. M. and D. G. McDonald. 1987. The physiology of acid/aluminium stress in trout. *Belgium J. Zoology.* 117(1):399-410.
10. Willhite, C. C., et al. 2014. Systematic review of potential health risks posed by pharmaceutical, occupational and consumer exposures to metallic and nanoscale aluminium, aluminium oxides, aluminium hydroxide and its soluble salts. *Critical Reviews in Toxicology.* 44:1-80.
11. APHA. 2005. Standard methods of chemical analysis of water and wastewater (21st edn). American Public Health Association, Washington D.C.
12. Geldenhuys, A. J., et al. 2003. An integrated limestone/lime process for partial sulphate removal. *J. Southern African Institute of Mining and Metallurgy.* 103:345-353.

13. BIS. 2012. Specification for drinking water. IS 10500:2012. Bureau of Indian Standards, New Delhi.
14. Brahaita, I. D., *et al.* 2017. The efficiency of limestone in neutralizing acid mine drainage – A laboratory study. *Carpathian J. Earth and Env. Sci.*, 12(2):347-356.
15. Skousen, J., *et al.* 2017. Review of passive systems for acid mine drainage treatment. *Mine Water and the Env.*, 36(1):133-153.
16. Pyrbot, W., *et al.* 2019. Neutralization of acid mine drainage contaminated water and ecorestoration of stream in a coal mining area of East Jaintia Hills, Meghalaya. *Mine Water and the Env.* <https://doi.org/10.1007/s10230-019-00601-9>.
17. Sun, Q., *et al.* 2000. Effects of armoring on limestone neutralization of AMD, Division of Plant and Soil Sci., West Virginia University. 1(21-24):21-27.
18. Ahamad, K. U. and M. Jawed. 2010. Kinetics, equilibrium and breakthrough studies for Fe (II) removal by wooden charcoal-A low-cost adsorbent. *Desalination*. 251(1-3):137-145.
19. Schutte, F. 2006. Handbook for the operation of water treatment works. Water Research Commission, Republic of South Africa. pp 1-242.
20. Al-Rawi, S. M. 2009. Introducing sand filter capping for turbidity removal for potable water treatment plants, Env. Research Center (ERC), Mosul, Iraq. *Int. J. Water Resour. and Env. Eng.*, 1(1):011-019.
21. Nemade, P. D., *et al.* 2009. Removal of iron, arsenic and coliform bacteria from water by novel constructed soil filter system. *Ecological Eng.*, 35(8):1152-1157.
22. Sayed, G. 2013. Treatability study of wastewater using activated carbon, sand filter and dual media filter. National Conference on Biodiversity: Status and challenges in conservation - 'FAVEO' 2013. Proceedings, pp 210-213.